

FORMING A COMPLETE RECORD OF THE PROCEEDINGS OF ALL PUBLIC COMPANIES.

[PRICE 6D.]

ROYAL CORNWALL POLYTECHNIC SOCIETY.
PREMIUMS FOR 1882, NOT CONFINED TO THE COUNTY.

A premium of Fifty Pounds, by the Rev. Canon Rogers, to the superintending engineer, who shall have been employed to erect a machine for raising miners from underground, in any Cornish mine.

A Gold Medal, by a society, for the most approved model, with drawings and explanations, of a merchant vessel, of from 200 to 300 tons—combining strength and sailing qualities.

A premium of Five Guinea, by W. Broad, Esq., for the model of a merchant ship, from 250 to 300 tons.

A premium of Ten Pounds, by W. T. Prout, Esq., M.P., for the best chemical or mechanical method of ventilating mines, that can be applied to the Cornish mines with advantage.

A premium of Ten Pounds, by T. J. A. Roberts, Esq., for the plan, or model, which obtains the above premium, provided it be accompanied by an estimate of expense as shall be approved of by the Judges.

A premium of Five Rupees, by John Taylor, Esq., F.R.S., for the most complete and accurate accounts of the quantity of water supplied to the houses, the number of bushels of coal consumed, and the duty performed by an engine, for a period of not less than six months.

A premium of Ten Pounds, by Henry English, Esq., for the best paper, containing an account of any methods or plans, practised in any other mining districts, advantageously applicable to the Cornish mines.

A premium of Ten Pounds, by Lieut. Colonel Sykes and Sir C. Lemon, Bart., for the best report on the agricultural condition of any union in the county of Cornwall.

A premium of Ten Pounds, by the society, for the best practical method of obtaining the circulating efforts, produced by solution of copper and other substances, in the food and incense water on the boilers, and other parts, of the steam engines used in Cornwall.

A premium of Three Guineas, by Charles Fox, Esq., for the best analysis of the air taken at the termination of a "cure" of two men, from the extremity of one mine level in granite, and of another in kyllan.

A premium of Twenty Pounds, by Major Leubus and Messrs. G. C. and R. W. Fox and Co., for the best method of extracting alkali in dung waste, and manure.

A premium of Five Pounds, by the society, for the best shaded drawing, from a plaster cast, after the antique.

A premium of Two Guineas, by Miss Fox and the society, for the best-drawn sketch-book from Nature.—Any one under eighteen may compete.

A premium of Five Pounds for the best drawings illustrative of the five orders of architecture.

A premium of Two Pounds for the best map of Asia, showing the tracks of the

A premium of One Pound for the best perspective outline, with illustrations. Charles Fox, Esq., offers to the society, as long as he continues a member of it, the sum of Five Pounds annually, to be distributed in the respective sums of Two Pounds, One Pound, Twelve Shillings, and Eight Shillings, in four several prizes, for the most and most correct maps of some one state, province, or Burgh, county, comprising not less than 144 square miles, or a portion of not less than 100 square degrees of some uncivilized region.

Much inconvenience having arisen from the late period at which articles have been sent in for competition, it is particularly requested that, on all future occasions, they shall be lodged at the Polytechnic Hall three or four days before the day of exhibition, whereby the merits of every article will be more particularly examined.

Any other information respecting the Society may be had on reference to the members of the committee, or the agents in the country, from whom the reports of the society may be obtained.

ROBERT HUNT, Secretary, Fulham.

AGENTS.—Falmouth, Mrs. Trahan; Tyne, Mrs. Heard; Penzance, Mr. Rodin; Alderney, Mr. R. Blos, jun.; Camborne, Mr. L. Newton; St. Austell, Mr. J. Drew; Richard, Mr. R. Estabrook; Lantwithel, Mr. R. White; Bodmin, Mr. Liddell; St. Ives, Mr. H. Carlyn. London—Hopkin and Marshall, Stationers' Hall, 61, Strand, 69, High Holborn.

ANDREW SMITH'S PATENT WIRE ROPES, for standing rigging, lightning conductors, strapping of barrels, mining, railway, and general purposes; about half the size and weight of common ropes, and at one cent. Testimonials to that effect, with specimens, may be seen, and every information obtained, at the office, 37, New-Broad-street, city, manufacturers, Mill-wall.

Matthew Owen	Newcastle-on-Tyne.
Joseph Southway	Plymouth.
John Thompson and Co.	Wigan.
J. Y. Tregenna	Truro.
Thomas Mosney and Son	Dublin.

Perrin and Dale. Winkler,
Cotton and Young. Belfast,
James Giblin and Co. Glasgow,
James Giblin. Leeds,
J. M. Scott, Cromwell's lane, High street. London.

ANDREW SMITH'S PATENT WIND ROSE

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The rope has been in use for standing rigging in her Majesty's Navy, and in a vast number of merchant vessels, for upwards of six years, and is giving the highest satisfaction; the rope is also employed in various mines and railways in different parts of the Kingdom.

TO THE MINING INTEREST OF CORNWALL.—SAVING
OF FIFTY PER CENT.—I. T. THURILLAN, Treas. Agent for ANDREW
WITTH'S PATENT WIRE ROPE for the county of Cornwall, feels it a particularly
satisfying duty, after nearly two years' highly satisfactory trial in the Cornish mines,
to report to all concerned, that the strength of the said trial has confirmed beyond a
doubt his best conviction, that a saving of fully one-half of the expense of ordinary

could be effected, and is now effecting, by the use of those ropes, composed of
 of fish bone, over those made of Russian hemp. — J. F. Tregelles sends the follow-
 ing testimonials from the many he has received :—

Wheat Mary, Leland, Oct. 10, 1841.
 Sir,—I have much pleasure in informing you that the 2-inch wire rope which you

Use the 10% of July and has fully recovered your communication, but I need
ed it to lift five tons and agree with perfect confidence.
I am, Sir, your's, most respectfully,
Mr. L. T. Freggier, Trans. (Signed) R. E. MURILL.
North Quebec Mines, 12th Feb., 1922.
Sir,— The 12 inch wire rope you send in three mine has been at work at our 25.

cyinders steam, who about two months drawing from oil to oil, between, is not at all the worse for wear—I have examined it daily, and nothing can be better. I shall be happy to show it to any person who may apply, or I can give it the best of evidence of the kind that has ever been obtained, and from the day it does, and from the day it ceases, used more by constantly used, and from a chain thrown out. The wearing of coal is full as hard. From the stealer

I am, Sir,
Yours faithfully,
JAMES WATSON, Engineer.

To Mr. J. F. Thompson, Town.

Price 75c. per Cwt.; First and Small Round 50c. per Cwt.
A. 10-lb. wire rope is a full size within rope, weighs about 7 lbs. per fathom, and
is all right here.
B. 10-lb. which weighs 7 lbs. per fathom, will do the work.

¹⁰ These wire mesh are now at work at the following mines - Thompson, The Big Creek, Conestoga, Whelan, Mass. 11-10-11, West, Northampton, Mass.

...and Washington, D. C. They are now in use at the Mayo Grapeshouse, Fresno, Calif., and Washington, D. C. — Also for grape vines, etc., at Fresno, Fresno, and many other places.

NEW PATENTS FOR FEBRUARY.

Charles Wye Williams, *engineer*, *Amersham*, for certain improvements in the making and mounting of *bricks*, *streets*, *foot*, and *other* *structures*.
 Benjamin Broom, *colliery* *engineer*, *Amersham*, for certain improvements in the construction and operation of *colliery* *engines*.
 John Higgs, *chemist*, *Amersham*, for certain improvements in obtaining motive power, by means of *colliery* *engines*, and *other* *structures*.
 Charles Thomas Holcombe, *iron* *merchant*, *Bankside*, *Southwark*, for certain improvements in the manufacture of *fuel*, and in obtaining products for such manufacture.
 John Leathwaite, *engineer*, *East* *Street*, *Manchester*, for improvements in steam engines and boilers.
 Thomas Plodell Grampson, *engineer*, *Lipson* *grove*, and John George Hadden, *civil* *engineer*, *Blountgate* *Street*, *London*, for improvements in steam engines and railway carriages.

IMPROVEMENTS IN STEAM-ENGINE BOILER AND FURNACE FLUES.

(Specification of the patent granted to Joseph Aney, of Gowerwall-road, Lambeth, engineer, for improvements in the construction of flues for steam-engine boiler and other furnaces.)

These improvements consist in making the flues around nearly the whole outer surface of the boiler, the outlet into the chimney being at a point below the water-line of the boiler; by which means, although the flues are above that line, the rush of the flame, heated gases, &c., will not be higher than the outlet into the chimney, all above that outlet being a stagnant heated atmosphere. Thus, any injury that the metal of the boiler, above the water-line, would sustain, if the rush of flame to the chimney was above that line, is prevented. The patentee claims the mode of constructing flues of steam-engine and other boiler furnaces, whereby the same are carried above the water-line of the boiler, and the outlet into the chimney is found below the water-line of the boiler, as above described.

IMPROVED PROCESS OF CASE-HARDENING IRON.

(Specification of the patent granted to Robert Roberts, of the township of Bradford, near Manchester, in the county of Lancashire, blacksmith, for a new method or process of case-hardening iron.)

This invention consists of a new mode of combining or uniting wrought-iron with cast-iron, by which wrought-iron, of any manufactured form or dimensions, may be readily case-hardened. In order to perform the operation of case-hardening iron, according to this process, an ordinary furnace is prepared, in which is placed a vessel or compartment, to contain coal or moist iron, heated and molten by the furnace, in such a state as it would be used for running or casting into moulds. In the same furnace is placed, in a separately heated chamber or compartment, the article of wrought-iron intended to be hardened, say (for the sake of illustration) the tyre of a wheel. The furnace may be of any form or dimensions, suitable to the shape and size of the wrought-iron to be hardened. The furnace being closed, and the cast-iron in a state of fusion, the wrought-iron tyre must, when at a complete and heat all over, be lowered into the fluid cast-iron in the vessel, chamber, or compartment, below, and gently rolled or turned round in it, when the cast-iron will completely case over the wrought-iron tyre with any required thickness of cast metal, from one sixteenth of an inch, more or less, which, upon being removed from the furnace, and immediately immersed in cold water, will be suitably case-hardened or steel.

It will be necessary to observe, that the iron, as it comes out of the furnace, should be so immersed in water, that its entire surface be covered by the water at the same time, in order to prevent any irregularities in the operation of the metal. Great elasticity and strength may also be given to hard or strong wrought-iron, thus case-hardened; and carriages and other springs may be made in this manner. The patentee claims the method or process of case-hardening iron, by casting, covering, or combining wrought-iron with cast-iron, as herein set forth.

IMPROVEMENTS IN STEAM-ENGINES.

(Abstract of specification granted to John Thomas Carr, of the town and county of Newmarket-on-Tyne, for improvements in steam engines.)

The steam-engine, constructed according to these improvements, has its piston-rod working through a stuffing-box, having the character of a universal joint. This stuffing-box is a spherical box, working in a ball upon the top of a box which slides to and fro by dovetailed joints in the steam cylinder cover. The piston rod is jointed to the piston and attached directly to the crank of the driving shaft; the universal jointed stuffing-box and the sliding box, conforming to its movement and vibrating backward and forward to the extent of its deviation from the perpendicular. The air-pump is placed immediately beneath the cylinder, and worked by its bucket rod, being attached to the under side of the steam piston, passing through a stuffing-box in the base of the steam cylinder. The valves are worked by the sliding-box before mentioned in the following manner:—A stud or pin projects from the front side of the box and works in the lower bush of a T-shaped lever, seated in the middle of the horizontal portion. At each extremity of the upper arm is fixed a pin, from one of which a connecting rod passes down to the valve rod. As the box slides backward and forward upon the cylinder, a rocking or oscillating motion is given to the lever, and the raising or depressing of the valve-rod effected. In order to reverse the motion of the engine, the valve-rod is merely shifted from the one arm to the other, which brings about the desired end. Another arrangement for reversing the motion of steam-engines, without altering the valves or gears, consists in placing an intermediate slide or post-piece between the D valve and the cylinder ports, by shifting of which the induction and exhaust passage becomes reversed.

For working the cold and hot water pumps, the following arrangement is adopted:—On the opposite end of the main shaft to that at which the crank is situated, the first motion wheel is keyed, and on the outer face of this wheel a pin is placed eccentrically to its axis; from this a connecting rod passes down to a triangular-shaped block of metal, working between two upright guides; to this block, the piston of the cold water pump, and the plunger of the hot water pump, are attached—the object of the weight being to counterbalance the piston, air pump, bucket, &c. In lieu of the ordinary governor, the patented engine employs the following apparatus:—A pulley is driven by means of a belt on the driving shaft, and upon the periphery of this pulley runs a smaller pulley, attached to a bell crank lever in connection with the throttle valve of the engine. As the regulated speed of working the two pulleys merely revolve in contact, but should that speed be exceeded, the small pulley is drawn forward by the increased velocity of the larger one, and the throttle-valve being acted upon, partially shuts off the steam.

The claim is, 1. To the method and arrangement for working the valves of steam-engines, by taking the motion directly from the movable stuffing-box of the piston rod, and conveying it to the valve-rod, and for the arrangement for reversing the motion of the engine. 2. To the method of reversing the motion of steam-engines, without altering the valves or gears. 3. To the method and arrangement for working the hot and cold water pumps, and for counterbalancing the weight of piston, air pump bucket, &c. 4. To the method and arrangement for regulating the motion of steam or other fluid engines by means of a frictional governor.

ACCIDENTS ON RAILWAYS.—From a Parliamentary return, we learn that, during the year 1841, the number of accidents on various railways, which arose from causes beyond the control of passengers, or, in other words, from carelessness and want of proper precautions on the part of the railway companies, or their servants, amounted to twenty-nine, by which twenty-five persons were killed, and seventy-two injured. During the same year there were thirty-six accidents attended with personal injury to individuals, owing to their own negligence or misconduct, by which seventeen persons were killed, and twenty injured; sixty accidents occurred which were attended with personal injury to servants of the company, under circumstances not involving danger to the public; by these accidents twenty-eight individuals were killed, and thirty-six injured. Thus there were altogether sixty-nine human beings whose lives were sacrificed by accidents on railways during the year 1841.

ANTHRAX AT GARNKILL.—At the sitting of the Paris Academy of Sciences, on the 11th instant, a communication was read from M. Combes, on the destruction of the malleable tube in the bore of the wall of Grenelle, in which he suggested that the phenomenon might be accounted for in the following manner:—that, by the stoppage of part of the tube by earthy matter, or anthrax, both above and below, a kind of vacuum had been produced, and the water, which still remained outside the tube, between the metal and the earth, had exerted so great a pressure, that the metal immediately gave way, and increased the force it was found to possess when exhausted.

THE MACHINERY.—Mr. Hall, of the Marble Works, Derby, has in his possession an interesting relic of the antiquarian world, a large small tooth, supposed to be of the machine, which has been recently found in the alluvium in the neighbourhood; the central parts of the tooth, which are in an excellent state of preservation, are perfectly unaltered, but the outer parts are not so much indicated; the whole is beautifully finished, and exhibits various degrees of hardness.

PROFESSOR HUMAN FAVES.—A paper was read of the Academy of Sciences, Paris, on a pretended human skull, coming from the cabinet of the late M. Tournefort, but without any indication of locality attached to it. It was found to have been all out of bone particles, which had been replaced by hydrated oxide of iron and manganese. It was concluded that the skull had fallen into some quarry in which there was a deposit of lignite, or wood-ash, and had there undergone the change in question.

LAW INTELLIGENCE.

IMPORTANT MINING CASE.

OXFORD CHANCERY—WEDNESDAY—MARCH 6.

MARSH V. ANGLESEY, LORD HATHERTON, AND ANOTHER.—This was a case of trover for the conversion of certain quantities of such iron, stone, and limestone, which the plaintiff alleged to be his property. The defendant justified the act of conversion by pleading a custom for the owners of cophold tenements in the manor of Cannock, in the county of Stafford, to take, from time to time, the coal, ironstone, and limestone found within their own copholds; and the affirmative of the issue raised on this plea being given for the defendant, his case was first laid before the jury.

It appeared that about six or eight years ago Lord Hatherton opened a mine on an estate called Romer hill, which is held by copy of court roll of the Manors of Anglesey, who is Lord of the Manor of Cannock. Upon being made acquainted with the fact of the establishment of this colliery, the Marquis of Anglesey addressed a reconvention to Lord Hatherton, stating that he had no right to work the mine without the plaintiff's consent. Lord Hatherton denied that any such consent on the part of the Lord of the Manor was necessary, and arrangements were accordingly made to try the right to dispute by bringing the present action. To prove an usage by the copholders for digging for coal and ironstone upon their cophold lands, evidence was given to show that about sixty years ago there had been workings in several cophold closes which were parcel of the manor, and that the coal had been openly sold by the copholders for their own profit, without any license from, or interference on the part of, the lord of the manor. It was shown also that coal and ironstone were frequently intermixed, and that it would very often not be possible to work one without the other; and further, that there were traces in one of the closes which indicated that iron had been smelted there. To corroborate this evidence the court rolls of the Manor of Cannock were put in, from which it appeared that in 1790 a surrender was made by one Barton and his wife, who were seized in fee of a cophold held of the lord of the manor, to another tenant for years, in which Barton reserved to himself the liberty to dig for coal, and to apply the coals suited to his own use. One or two other instances of a similar reservation were also produced. No evidence, however, was given as to usage by the copholders to take limestone. To meet the case thus presented on the part of the defendant, the plaintiff put in a grant from the Crown of the Manor of Cannock to the ancestor of Lord Anglesey, dated 30th Elizabeth, reciting several leases previously made by the Crown, in which the right to take minerals was made the subject of demise; and an agreement in 1605, made between William Lord Paget, then Lord of the Manor of Cannock, and several copholders of Cannock, Longdon, and Heywood, through one of whom Lord Hatherton claimed the Romer hill estate. This agreement, which was followed up and confirmed by a decree of the Court of Chancery, provided, in consideration of 1500*l.* paid to Lord Paget, that as disputes had arisen between the lord and the tenants of the manor, the customs of the manor should be afterwards taken to be such as they were set out in the deed, and among them was a custom for the copholders to cut timber on their cophold lands; and also to dig for clay, peat, sand, &c., on the lord's waste; but no mention was made in the agreement of any custom for the copholders to take the minerals found on their cophold tenements. The agreement, however, contained a proviso, that as "some matter or point of custom and therein be misremembered or expressed might hereafter come in question," such matters should be referred to an indifferent jury of fourteen or sixteen copholders; and it was contended by the defendant's counsel that the language of this proviso showed that the agreement was not intended to embrace all the existing customs of the manor, but only such as had been lately in dispute.

Mr. Justice CRESSWELL, in summing up the evidence, told the jury that if they were satisfied with the evidence of usage given by the defendant, and considered that it was not contradicted or explained away by the plaintiff's case, they might then fairly infer that the usage shown to have prevailed in modern times had existed at an earlier period. Most of the regulations contained in the agreement of 1605 related to incidents of tenure, but it was a matter for their consideration whether it was likely that the copholders, when they were paying money to the lord for the admission, were likely to forego any of their rights. The custom to cut timber was expressly mentioned, and when they were describing this sort of right, it was natural to expect that they would have specified all rights of that nature. It had been shown that the right to dig for coal and ironstone was a valuable one, and it was strange that when a custom to take turf, peat, clay, and so on, upon the lord's waste was stated in the deed, there should be no record of a custom for the copholders to take coal on their own land. Still it was possible that all parties might have been agreed at the time upon the right of the copholders to dig for coal, and they would not fail to remember that the exercise of the right was a clear and tangible thing, while the construction of this instrument was open to some doubt.

The jury retired to consider their verdict, and, after half an hour's deliberation, gave it for the plaintiff.

THE NORTHERN RAILWAYS' DISPUTE.

COURT OF CHANCERY—MARCH 6.

THE CLARENCE RAILWAYS, THE GREAT NORTH OF ENGLAND AND HARTFORD RAILWAYS.—The arguments on this appeal from the decision of the Vice-Chancellor (see JOURNAL of 19th ult.) were commenced on Tuesday, and concluded on Wednesday; in the course of the case, the Lord Chancellor having observed that he agreed with the Vice-Chancellor in his construction of the Acts of Parliament, but had doubts on the subject of the mischief arising from the agreements with Mr. Williamson, said, at the conclusion, that he should take time to consider his judgment.—(It is expected the case will be disposed of this day, Saturday.)

BAHIA STEAM NAVIGATION COMPANY.

VICE-CHANCELLOR'S COURT—MARCH 10-11.

HENSON V. HEATHCOTE.—The court was occupied the whole of the day in hearing the commencement of the arguments in this case; the points in dispute relate to the purchase of a vessel by the Bahia Steam Navigation Company.—The arguments were resumed, and concluded yesterday, when his HONOUR directed, as he had intimated he should do, that inquiries should be made in the Master's office, relating to the question in whom the vessels, the City of Kingston, the Star of the South, the Pearl, and the Pelican, were vested. With regard to the ship *Norwood*, his HONOUR said, that, as it appeared Mr. Heathcote was the purchaser of the same, and for which he paid 1340*l.*, and that he sold the same to this company for 1300*l.*, that he received 15*l.* as a commission on the sale; that he was allowed the sum of two guineas "carriage money" actually paid to the broker; and, moreover, that he had charged, and been allowed, the costs of the bill of sale, it was utterly impossible, in such a state of circumstances, not to see that Mr. Heathcote had made the purchase for 1340*l.* as agent for and on behalf of the company, and he should declare accordingly. His HONOUR, however, thought it best for Mr. Heathcote to say, that, in all probability, the charges of commission would have been proper and reasonable, if charged on behalf of a stranger; but, considering that he was one of six directors, and, as such, entitled to a certain amount of remuneration, he ought not to be allowed to make any further profit of the performance of his duties. If a "ship's husband" was necessary, it was obvious he would be the servant of the directors and require their superintendence, and which superintendence they were paid to give. Here one of these directors became the "ship's husband," and it had been argued that the court could allow him his charges. Such a course was utterly incompatible with the principles of a court of equity, and if allowed would be the greatest possible encouragement to fraud. Mr. Heathcote must, therefore, pay all the costs of the suit occasioned by the charge or allowance of such costs, and full inquiries must be made relating to those charges in the Master's office.

The case is to stand in the minutes to be spoken of on Thursday next.

IMPORTANT TO SHAREHOLDERS.—THE AGRICULTURAL BANK.—A notice has been given at the Agricultural Bank Association of Mr. Green, of Belfast, and Mr. Adams, of Portlough, as shareholders of the Agricultural Bank; the notice set up was, that they had not signed the deed of partnership. This verdict establishes as fact, that the record at the Stamp-office is sufficient evidence against a shareholder.

WOOD PATENT.—A lecture was delivered, on Wednesday week, at the Royal Victoria Gallery, Manchester, by Mr. A. B. Birklin, on the subject of paving streets and highways with blocks of wood, and the advantages from that of stone. The lecturer, in tracing the introduction of wood paving into this country, gave the merit of the idea to Mr. John Finlayson, who, in 1823, published an essay on the subject in the *London Journal of Science*, in which he endeavoured to prove (after having made experiments) that the durability of wood was much greater than granite; but the merit of carrying into effect this new system of paving, Mr. Birklin awarded exclusively to Mr. David Wood, of London, he having secured a patent in May, 1838, embracing pavements of wood of every form and shape, and, moreover, to erect his specifications within the legal time, which informally was taken advantage of, and was the means of bringing many competitors into the field. Of all the plans, however, that have been introduced, the lecturer considered that of Mr. Wood as decidedly the best, being formed upon strictly mechanical principles, while the others were evidently mere imitations, to evade his patent.

CHANCERY.—A series of popular lectures on chemistry are in the course of delivery by Dr. Reid, at Exeter Hall, London; the first lecture was given on Wednesday last, which was mostly of an elementary character.

ON THE NATURE AND PROPERTIES OF CARBON.

Mr. Winkler has just published a series of interesting and valuable lectures on Chemistry, at the request of the Royal Society of Medicine, Birmingham, the principal points embraced in which we shall occasionally quote to our columns, as, upon many of them, we have been delivered to the order of the lectures in which they were presented, may have been delivered:—The fourth was chiefly occupied with a consideration of the nature and properties of carbon—an elementary body (said the lecturer) which formed, in fact, the basis of animal and vegetable substances, and entered largely into the composition of coal and other bituminous matters. Carbon was also found abundantly in many mineral rocks; limestone, chalk, and marble contained it in large quantities, but in these the carbon combined with oxygen. Although very universally distributed throughout Nature, carbon was seldom found in a pure uncombined state. In its pure form carbon constituted the diamond. If, for instance, they took any kind of wood, and exposed it to a red heat in a close vessel, the volatile parts would escape, and carbon or charcoal would remain behind; this, deprived of its earthy matter, formed the substance of which the diamond was composed, the diamond being nothing more or less than carbon in a crystalline form. Chemical investigation and experiment had placed the identity of charcoal with the diamond beyond dispute: Charcoal possessed some very extraordinary properties; amongst these was the property of resisting putrefaction; it was not easily acted upon by other elementary bodies, and, therefore, it retained its pure uncombined state, under ordinary circumstances, for a great length of time. The ancients appeared to be well acquainted with this peculiarity of carbon; and it was customary with the Romans, before driving piles into the earth for the construction of bridges and similar works, to char the end of the wood by burning, and it was found, after the lapse of a thousand years, that the piles so prepared underwent very little change. Charcoal also possessed powerful antiseptic properties, meat being preserved from putrefaction by washing it with water impregnated with powdered charcoal, and water itself of the foulest kind being readily purified by passing it through alternate layers of sand and charcoal. This preservative property was doubtless communicated to charcoal from its affinity for gaseous bodies, and these were given out freely by animal and vegetable substances in a state of decomposition, it having been ascertained that newly made charcoal would absorb sixty times its own volume of some of the gases. It was difficult to account for the circumstance of charcoal absorbing an amount of gas so much greater than its own bulk; but all they could say was that such was the fact. The lecturer proceeded to notice some of the chemical combinations of charcoal and oxygen. Amongst these was carbonic acid, which possessed highly poisonous properties, and could be made from many substances containing carbon and oxygen. It was generally made from sugar, and from its crystalline form was sometimes mistaken for epson salts. It might, however, be easily detected by its peculiar acid taste. In case of its being taken into the stomach by accident or otherwise, its poisonous property might be effectually counteracted by swallowing chalk diffused in water, or by taking magnesia. In the absence of these remedies the scrapings of white wash, from the wall or ceiling of a room, mixed with water, would be found to act as an effectual antidote. The great abundance in which carbon was found in coal led the lecturer to glance at the theory of the formation of that valuable mineral, and to recapitulate some of the leading facts in favour of its vegetable origin. By heating coal in a retort they obtained precisely the same elements as were produced by decomposed wood or any other vegetable substance, but in different proportions. The lecturer afterwards noticed the valuable invention of Sir Humphrey Davy for preventing explosions in mines, and explained the principle of the safety-lamp, concluding with some interesting observations on the first introduction of coal into London, in 1275, when a medical council was held to report on its influence on health, which resulted in the issuing of a proclamation by Edward I. prohibiting its use within the bounds of the city.

MINERAL RESOURCES OF THE INDIAN PENINSULA.

At a late meeting of the Asiatic Society, a paper was read by Lieutenant Newbold, "On the Copper Districts of the Southern Mahratta Country and Nellore," which was understood to be the first of a series of papers preparing by that gentleman on the mineral resources of the Peninsula, from which he has brought an interesting collection of ores and other minerals, selected during a residence in the country, to which he is shortly about to return. The origin of the extensive, but now almost neglected, copper mines of the peninsula, appears to be lost in vague traditions; they are said to have been worked under the Vijayanagar Dynasty, within whose empire they were situated, and by the Mogul conquerors, who supplanted it; but we have no further evidence of the progress followed, or of the produce obtained than can be gathered from the magnitude of the excavations, and the immense mounds of slag and scoria, now covered with vegetation, which point out the situation of the smelting-furnaces. We may infer, that, however rude they might have been, the native processes were effectual in separating the metal from its ore, as few traces of copper remain. About forty-seven years ago the copper mines of Nellore were brought to the notice of the British Government by Dr. Hays; but no result appears to have followed. A speculation entered into about eight years ago failed for want of capital, and some mismanagement. Some about two years ago they were again taken in hand by J. Ochterlony, Esq., who has established a skillful Cornish miner on the spot. The country in which these mines are situated is an undulating plain, extending from the foot of the Ghats to the Bay of Bengal, and studded with a few detached, round-topped hills of granite, mica, and hornblende schist, rarely a height of more than 250 feet above the adjacent country. Much of it is barren and neglected; and large tracts are almost without a blade of grass. Hardy mimosa are the only vegetable objects in many extensive tracts, and everywhere the vegetation is stunted. The rocks comprise mica, granite, and hornblende schist, highly gneissiferous. Granite is rarely visible, except in veins traversing these rocks, but basaltic dykes are very common. Much of the paper was taken up with details of particular mines, which will not admit of abridgment. Analysis of some of the ores have been made to England by Dr. Thompson, and in India by the talented Pioneer. The latter gentleman found the ore submitted to him to be composed of sulphur and carbonate, and to contain 90 per cent. of pure metal. The green carbonate associated with quartz gave above 30 per cent.; and other specimens contained quantities varying from 40 to 60 per cent. It would appear that there are great facilities for the successful working of many of these mines. The population of the country is very industrious. The natives of the district about Gurnahy, or Garipahy, which may be called the capital of the mining district, are of the Upper caste, whose business is well or tank digging. The common trees, near the coast, furnish abundance of coal for roasting; and charcoal is sold at the small end of three annas for one anna. (If the Madras anna is meant, this will be about forty pounds for a penny.) Ramapatnam, on the coast, less than forty miles distant, offers an eligible port for shipping the produce; and it has been suggested, that the river which runs by Gurnahy ponds, might be made navigable, though this is doubted by Lieutenant Newbold. On the whole, it may be considered that these districts afford a fair prospect of giving employment to a large proportion of the population of the country, and of contributing to the resources of Great Britain.

At the conclusion of the paper, it was remarked by Colonel Sykes, that India afforded a rich, unexplored field for such researches as had been so ably exhibited by Captain Newbold. It was known that gold existed in the streams of the Nellore district, and in the Dhyra dam; but he was not aware of any steps having been taken to trace up the auriferous streams to their sources. He observed, also, as an instance of how little had been formerly known of our resources in India, that fifty-seven localities were now ascertained to produce coal, whereas, but a few years ago, the existence of that mineral was wholly unknown.

EARTHQUAKE IN CORNWALL.—At a late meeting of the Royal Cornwall Polytechnic Society, Mr. Robert Hunt (the secretary) gave the following particulars of the results of his inquiries into the extent of the disturbance of the earth which was experienced in this locality on the morning of February 17th. It does not appear that the shock was felt on the northern side of Carr Marsh or Carr Brea. At Wreath Virgin, one of the Consolidated Mines, on the junction of the hills with the granite, and at Faldry, a part of the United Mines, similarly situated, the shock was sensibly felt 150 fathoms from the surface, but not at all to the eastward of these mines. On the south side of Carr Marsh a very sensible tremor is described, and the men at all depths in Treowen felt the shock. In the southern parts of Camborne it was also felt, and at South Wreath Basset the hillside fell left his work and came to the surface. In St. Ives and Camborne the effects appear to have been greater than in any other parts, particularly the latter parish, from which that may be regarded as the centre of the disturbing force. Along the line extending from about half a mile east of Carr Marsh and the same distance west of Carr Brea, towards the south the noise was heard, and in many places a tremor felt; under these lines (Perran, Falmouth, and Helston, are conspicuous for the very decided manner in which they experienced the shock. At Wreath Var, at the 150 fathoms level, and below the 200 fathoms, the shock was most distinctly felt by the men. We still hear of it towards the Lizard, but with a very modified action. At Portrevere the shock was scarcely felt, and but very slightly at Colwall and Cremack. It is curious to observe that the great line of action is confined to the granite range running nearly north-east and south-west, and that although the influence of the shock extended into the hills and surrounding districts, it was gradually weakened, and soon lost entirely.

GEOLOGY.—A NEW SYSTEM OF PHILOSOPHY.—No. IX.

BY HENRY GRAMM MONTAGUE, ESQ.

PHENOMENA OF THE DESERTS.

The Book of Nature is opened—the sun of truth, dispersing the clouds of superstition and folly, bursts forth in its meridian blaze, dawning the eyes of the unlearned, and confounding the wisdom of the wise. Who would seek shelter from its glorious beams beneath the dark and cumbersome mysticisms of the past? Is not knowledge power? The buildings of ages totter to their fall—opinions, theories, and systems, crumble into dust—and the proudest monuments of old yield to the talisman of knowledge as mind awakens to its strength.

Hearken unto me ye children of the earth—come hither! drink at the fountain of Nature, and be wise. Wisdom is a plenteous draught to the thirsting soul; like the effulgent glory of the sun, giving light and life, where light and life were previously unknown. Inhabiting the hallowed temple of the soul, her all-searching eye wanders o'er the earth—through the ocean depths—through the immensity of space above and around us—gathering food for meditation and for wonder, gathering material upon which to erect the edifice of greatness and of immortality. Who will listen to the voice of wisdom! striking upon the senses as chords of harmony awakened by perfection? Who will gaze upon her charms! as the sun shining in his strength—the fair array of all that's bright and beautiful? Who can describe her form? Omniscient, omnipresent, the source of all power, the fountain of all greatness, the glory of intelligence! Behold her enthroned in the temple of Truth, enrobed in the pure garments of light and immortality! On her right hand stand the attributes of good and of evil, of life and of death—on her left is the image of the past—before her is spread the volume of reprobation, open to the view of all, and disclosing the past, the present, and the future. Pointing to the fair unspotted volume, she exclaims—"How long will mortal man prefer darkness to light? How long will he refuse the sweet refreshing stream to revel in polluted waters? Perishing from want, will he refuse food? Parched with thirst, will he pass the cup unstated from his lips?"

Berthollet is of opinion, that, within certain limits, bodies have a tendency to unite in every proportion; and that combination is never definite and invariable, except when rendered so by the operation of modifying causes, such as cohesion, insolubility, elasticity, quantity of matter, &c. In the phenomena of production are evidenced an endless variety of changes, the act of change being almost wholly unknown to us, motion in and through all governing matter in its disposition and movements. Every compound, whether it be elementary or proceeding therefrom, has certain capacities of action and powers of retention; the same varying with the disposition of its parts and qualities, the result being, at all times, determined by the body in action, the limits of cohesive power being the limits of retention; the same, or product, must, of necessity, be definite, when matter is united with matter in definite admixtures; but the sum of each, in uniting, depends on the nature and qualities of the bodies, and the action manifest in the act of union, this truth being demonstrated by numerous experiments of the practical chemist, and more extensively so by Nature; for example, atmospheric air is an union of compounds in definite proportions, the elements of its composition being oxygen and azote; each of these elements has varying powers, the maximum and minimum of which are, their condensed and expanded volume. In their expanded volume oxygen has the capacity, but not the power, of action—azote has the capacity, but not the power, of retention; thus, the one blends with the other, as muriatic acid blends with water; but, as oxygen concentrates its volumes, so it attains the power of action, and, as azote concentrates, so it attains the capacity of retention of the oxygenic body, to the extent of those powers, and within those limits; thus these bodies may be united in variable, but not in every, proportion, and each combination is definite and invariable when thus mathematically united, the union being a perfect result, but, in common with all other results, the subject of incessant change of separation and re-combination. In union, these elements being compounds, having one common base, the results proceeding from admixture with other bodies, are still more uncertain, the same depending on the conjoint powers of action and retention of these bodies, and the elements of which they are composed. Experimentalists, by uniting many vegetable acids, and causing them to pass through the fire, produce the result, acetic acid. Nature, in the decomposition of ten thousand organic animal and vegetable bodies, gives the combined results, carbon and lime, but, to produce the one or the other, certain action is necessary, and upon this action alone the result depends.

Chemists, under certain combinations, produce potassium, aluminum, and other, as termed, simple bodies; but in Nature they are produced, or not produced, as the accidents of circumstances may determine; for instance, potash is not found in wood under the form of coal, although it is the acknowledged property of wood; but, say the chemists, in the chemical change of the material the matter has been abstracted therefrom; such is not, cannot be, the case, for, was it so it would be in the over or underlying beds of other matters, which it is not; and, again, potash is a result of compulsory union of bodies by fire, as is also aluminum—the one or the other are never known in Nature, the same may be said of calcium, the elements which in union form the product are in innumerable organic bodies, but this product is not invariably developed in aggregates composed of these bodies; and, again, of carbonaceous bodies, it is said that charcoal and diamond are both pure carbon, but such is not the case, for in the first place carbon of itself is a compound product, and its development depends upon the accident of circumstance, for if produced by, and in, oceanic animal matter, its constituents may assume the form and properties of silica; if produced by terrestrial animal and vegetable matter, its constituents may assume the form of alumina or potash; again, the animal matter becomes bitumen, silica, earth, or carbon, as the accidents of outward influences or combinations may determine; and the same may be said of vegetable matter, which, under varying influences, passes into bitumen, silica, earth, or carbon.

Admitting that a compound substance, so long as it retains its characteristic properties, always consists of the same elements united together in fixed proportions, such being in truth the law of affinity governing combinations, still it must be acknowledged, that however true the compound result may be, the maintenance of its body and compound properties is, under all influences, very uncertain, for the same law of forces called into operation by accident or by design dissolves the union, and produces other results equally fleeting and inconstant, and depending, in like manner, for the maintenance of form upon absolute exemption from disturbing causes; thus, for instance, the elements of the human body form, in union with each other, numerous results, and the vital principle uniting the whole as one grand result, the phenomena of the living body is produced; but so long as life continues, the several compound substances entering into, and becoming portions of, the body, are but the results of a moment, constantly changing in their disposition, quantities, and qualities; and when by death the general link is broken to pieces, these several compounds exhibit their characteristic properties only so long as they are enabled to offer passive resistance to elements acting upon them from without, being, perished at all times to enter into new combinations in obedience to the law of force of affinity. Wood, on the abstraction of its hydrogen, may be converted into dust; with hydrogen and nitrogen this may be converted into vegetable earth; by the addition of water it becomes clay; by the abstraction of hydrogen and accession of oxygen and nitrogen it becomes rock; again, by the abstraction of nitrogen, carbon, sulphuric acid, or other compound through atmospheric or volcanic action, the rock decomposes, and the characteristic properties of the body may be at once, and for ever, obliterated, giving place to others in the recombination of elementary bodies with each other, the change, and the nature of the change, depending entirely on the accidents of combination, and the separate or conjoint powers of the bodies coming into contact.

In my last article I spoke of the origin of species of porphyry common to the Deserts, and manifestly of recent formation, the universal compound superstratum is formed in these localities, in a manner peculiarly similar to porphyry, magnesia being the chief constituent, and silica nearly in the same proportion, building in union carbonate acid, lime, ironoxide, and other compounds, as it is to be seen in large irregular masses disposed within the calcareous strata, and sometimes resting on a bed of sand, flinted, and forming in like manner, with the marbles, sometimes resting with granite, or being capped with granite. I now come to one of the most important compounds known to man, under the form of rock, and upon which several systems of geology have been founded.

GRANITE is a term arbitrarily applied to certain aggregate masses of matter, consisting of grains, or concrete particles, of felspar, quartz, and

mica, intimately united by a common base, or cement, which is in general silica or alumina, or both united. These several materials vary in their admixtures with each other, and meet in union, as the chances of local action may determine; felspar in general predominates, and mica is the least considerable ingredient of the rock; in some species quartz is wanting, in others the mica; and, again, the material passes, by transition, into sandstone, gneiss, and other rocks. The crystals also differ in colour and magnitude of parts, and also in the arrangement of the particles with each other; some granites resolve into sand, others into clay felspar with clay. The general outline of appearance is also variable; thus, some granites are of loose texture—are formed in hills and heaps, of a rounded form; others present the figure of limestone ranges, having precipitous cliffs; others are in beds and veins; again, some are stratified, others compact, and highly indurated. It often occurs in masses, many miles in extent, surrounded by gneiss, mica slate, and clay slate, and so connected with these rocks that the whole may be considered as one grand result of crystallisation, the whole being of contemporaneous formation. All the different rocks, erroneously termed primary, as granite, gneiss, porphyry, &c., alternate, and pass into each other.

Quartz, mica, felspar, and hornblende, are the chief materials of sandstones, shales, and clays; and the proximate principles, or compounds, of which the above compounds consist, are silica, carbon, lime, magnesia, alumina, potash, and soda; these being also the chief ingredients of all matter in aggregate of the earth, and which, as one whole, is termed earth. Granite, in common with all other rock, is developed in an interminable variety of combinations, the causes of effects being as variable.

Granitic rock is, by modern philosophers, termed primary, as being often the nucleus of a chain of hills or mountains; it is assumed to be volcanic, because it is composed of crystalline aggregates, united by one common base; it is assumed to be wholly destitute of organic remains, because, unlike some rocks, animal remains cannot be distinguished therein; but the above inferences are at variance with facts derived from observation, and, although they constitute the bases of modern systems, they are, nevertheless, false, and inconsistent with the operations of Nature. Granitic rock may have its origin in depositions of oceanic and terrestrial animal and vegetable matter in union, or a local portion of the ocean bed, wholly composed of oceanic matter, of calcareous qualities, may receive, on its surface, terrestrial matter, and, from this overlying matter, the elements which, in union with its own, constitute the material of granite, are derived. In the first instance, the matter, in its disintegrated state, may be a recent formation—and, consequently, in its crystalline state, it must also be a recent formation. In the second instance, the lower bed, which is the recipient of terrestrial matter, may claim priority of origin, but, in order to form granite, it is necessary that the terrestrial matter blend with the oceanic, the one and the other being conjoint proximate causes of production, as granite; thus, neither the one nor the other can be termed primary. Rivers convey alumina, potash, and oxide of iron, into the ocean; and these compound principles enter into the ocean bed, whereon they are deposited, and unite, with other terrestrial matters, with the matter constituting the ocean bed; in the lapse of time, in the revolutions of ages, the sea becomes dry land, and the united material is then the subject of atmospheric influences, and chemical action induced, the ultimate result being granite. The causes of production are—first, local production of oceanic and terrestrial genera and species; secondly, the union of their bodies in death; and, eventually, the consolidation of the whole by chemical action.

Granite is ever formed by heat, but never by heat of fusion; the disintegrated mass, exposed to excess of atmospheric heat for a succession of ages, must, of necessity, in common with all things else, undergo a change in its parts and qualities; by atmospheric heat, the hydrogen is carried off—the carbonate of lime, called into action, regulates the form of the crystal, and the silica and alumina are the cement to the whole, either together or separately. The size, texture, and material of the crystal, depending on the union of matter with matter, the action consequent thereon, and the lateral pressure, affecting the whole, in the consequent expansion of parts. The material of which granite is composed, being, of necessity, variable in composition and character, the result of union, as granite, is also ever variable; thus it is, no two rocks of this species can be found alike, for some granitic rocks have large and well-defined crystals—others have small; some are regular in their grain or fracture—others are irregular; some are triple compounds—others are quadruple; some are highly indurated, and, like the Egyptian granite, enormous—others are but slightly adhesive, and crumbling to the touch; some have certain definite compound admixtures—others are aggregates of many aggregates, differing in material from each other, although united by one common base; again, granite passes, by transition, into all other rocks.

The schistose rocks, so often found capping granite, or in contact with it, are at all times combinations of terrestrial and oceanic earths, and, in general, are found to consist of silica, alumina, magnesia, peroxide of iron, oxide of manganese, potash, carbon, sulphur, alumina, water, and volatile products; they are invariably sedimentary depositions, carried into the troughs and valleys of seas, or lakes, by the force of rivers and running streams, and therein deposited in the line of action, without reference to the bed on which they rest; but, if this bed be of decomposed matters, it becomes the recipient of compounds from the overlying stratum; thus, in addition to its primary oceanic qualities, it receives alumina and potash, the uniting characteristics of terrestrial matters. Sometimes the sedimentary depositions, produced, and produced by the action of running streams, contain, without reference to the bed on which they repose, all the elements of granite, and, under all circumstances, granite is formed by sedimentary deposition.

In the Deserts, the terrestrial matters are carried into the sands and calcareous beds by the fishes, by rivers, or by percolation from strata to strata; the waters and atmospheric heat acting conjointly on the decomposed mass, a new arrangement takes place in its atomic particles and aggregates, and the whole assumes the form of gravel, in which the compounds of the granite are severally, and independent of each other, produced, and, in this state, cohesion takes place—at first, simply as conglomerate, but, eventually, in the lapse of ages, the result is granite; if the waters are abundant, then it is the granite becomes almost wholly composed of quartz. In the Deserts, the blue line, as it is termed, develops a series of changes into trap, and from thence into granite, and, in all these changes, the internal character of the mass is precisely similar, the arrangement of its aggregates, and its final development depending upon local influences. The recent granites of the Deserts, in numerous instances, pass by imperceptible shades, into sandstone, and have been localised, sometimes, being sometimes but slightly adhesive; in Egypt, the several rocks pass so much into each other as to defy classification; in India, I have observed it exhibiting its enormous craters.

The most conclusive evidence that granite is produced from secondary causes, affecting primary matter, and that it is a combination of very recent formation, are the mineral aggregates induced within its matrix, or encasing into its composition. Many of the granites contain iron and tin in abundance, these metals forming sometimes full three-fourths of the bulk of aggregate; for the production of the one or the other of these metals, atmospheric action is absolutely necessary, and their combined state so distinctly negative volcanic action, more particularly so as copper, the presence of which is necessary for the production of metals, is at all times abundantly abundant to granite beds. All the metals are produced by, and in the union of, oceanic and terrestrial matters, and by electro-chemical action—and this action cannot be sufficiently induced by internal heat, granite, also, of necessity, contains a vast portion of atmospheric air, and more particularly ammonia, which is an acknowledged volcanic product; it also contains sulphuretted silver, copper, lead, zinc, arsenic, antimony, uranium, tungsten, and cobalt, all of which compounds are positively indurated with the granitic matrix, as are also arsenic, garnet, blende, stannic acid, beryl, corundum, actinolite, chlorite, talc, compound felspar, stearite, and numerous other compounds, many of which could not have been preserved their primary form and qualities had they been induced in a fused mass of matter, such as it is assumed to have been in common with lava. Again, in its disposition, it is equally variable and inconsistent, as are other mineral bodies, being above or below the strata in solitary clumps, in hills, or mountain craters, as the chances of localising may determine, being distributed in and over the surface of the earth in like manner with limestone and sandstone, without order or form defined. In Egypt it is found resting on a bed of clay, in undisturbed masses; it is seen in its several stages of production varying in its composition and character, and, in the ultimate result, being more conglomerate, of fine granular texture, approximating to crystalline limestone, of combined crystalline, and, at

times, approaching porphyry in beauty and comeliness. Again, its colours are infinitely varied, the same depending upon the accidents of combination.

In England, as in the lower granite veins of Babia, and in some of the Cornish mines, it is found in its soft state, the same hardening upon exposure to the atmosphere; it forms beds and veins when deposited over irregular surfaces, and is, by this means, introduced into fissures of stratum previously indurated.

Granite is sometimes almost wholly composed of quartz, which also enters largely into the composition of gneiss, mica, schist, and other rocks termed primary; and is sometimes, as previously observed by me, the material filling up large fissures, or intersecting the rocks in veins, as particularly exemplified in limestone, in serpentine, and in argillaceous schist, its degree of purity depending on the accidents of association. It unites, in its composition, hornblende, which is another chief component of many rocks, being a compound of silica, alumina, lime, magnesia, and water, with a trace of potash, the alumina and potash denoting the presence of a certain portion of terrestrial matter; united as hornblende, they enter largely into the composition of mountain rocks, syenite, and greenstone, but do not, of necessity, form a part of granite, as is exemplified in the Egyptian granites. Mica also enters largely into the composition of rocks and earths; it occurs with quartz and felspar, in granite and gneiss, but does not, of necessity, form a constituent of granite, although it denotes some species of this rock; it is a characteristic of the earths, being widely disseminated through the terrestrial beds; its chief constituents are silica, alumina, oxide of iron and manganese, and potash, and it is produced invariably by terrestrial influences; silica and iron are common to both elements, but the potash and alumina mark terrestrial action. Felspar also enters largely into the composition of granitic and other rocks, and is abundantly developed in the upper crust of the earth; its constituents are in prismatic felspar, silica, alumina, potash, and oxide of iron.

In drawing conclusions by the rules of analysis, as well as by observation, it is evident that the base of the mineral body, termed granite, is oceanic; the matters introduced in varying quantities, and which give its peculiar characteristics, are terrestrial, and the ultimate result, as a crystalline compound, is induced by atmospheric action, the aerial volatiles entering into, and becoming identified with, its consolidated state. That, in many instances, it protrudes through the overlying beds is an unquestionable fact, but the cause of effects are demonstrated by Nature in the present day, the submarine hills and mountain chains increasing within the waters, and consolidated or calcareous, retaining their respective quantities and qualities, until they become the recipients of oceanic products, when the extraneous matters from the overlying bed enter into, and become identified with, the primary base on which they rest. In proof of this, we have only to extend our views to those localities affected by the action of tropical rivers; the stream of the Ganges, during the rainy season, holds in suspension nearly one-fourth of its volume in mud, the same being composed of terrestrial and oceanic products; this vast accumulation is diffused over the bed of the Indian Ocean, including a space of some hundred miles, but principally in the line of tidal influence; thus, to the more distant points, a periodical deposit of terrestrial matters forms on the ocean bed, and, where the submarine hills divide these currents, the deposited matter envelopes their base, and sometimes blends with their material; the beds of seas fill up, by periodical depositions, or by disturbing causes, but the underlying bed, forming the nucleus of the group, cannot become crystalline until it is exposed to atmospheric influence, not, of necessity, by direct exposure, but by electro-chemical action manifestly proceeding therefrom.

Let the unbiased observer look at many of the granitic formations, they preserve the like direction and the like perpendicular faces with limestone, their sharp angles being preserved entire through a succession of ages, there may or may not be marks of disturbance in and around them, but in general there are not, the sedimentary deposits being in their primary state, and evidently affecting the material, and affected by it; thus, when granite is in contact with limestone, the parts immediately in contact become more highly crystalline; this geologists would impute to the heat of the granite mass, but will analytical chemists say so much? Will they tell us that quartz is produced by immediate contact with heat? that a fused mass of matter can be protruded through numerous overlying beds, without disturbing those beds, and assume a form such as the granite mountain chain presents? that proceeding, as is supposed, from one common fountain—the womb of the earth—it should be so dissimilar in its compounds and qualities, and still retain compounds which are of inflammable qualities, without these compounds being injured? I cannot believe it possible.

It is said that fossil organic bodies have never been found in granite; such is not the fact, they have been found in the Aberdeen and other granites in the finest state of preservation, and their presence is equally manifest in the quartz nodules of granite, as it is in the carbonates of lime.

LECTURES ON GEOLOGY, BY PROFESSOR PHILLIPS.—The first of a course of six lectures, "On the Fundamental Truths of Geology, with Theoretical Inferences and Practical Applications," was delivered by John Phillips, Esq., F.R.S., and G.S., in the Theatre of the Royal Manchester Institution, yesterday week; we shall publish in our next a pretty full report of the introductory lecture, and continue them in consecutive numbers, until the whole, which we have reason to believe will be of great interest, shall be placed before our readers.

CREATION OF LIFE BY THE AGENCY OF ELECTRICITY.—At a recent meeting of the Entomological Society, the subject of Mr. Graham's production of insects by the voltaic battery, which made some stir from its introduction before the members of the British Association, at their meeting in Bristol, in 1846, was revived by Mr. Newport, who stated that a friend of his, at Sandwich, had produced the *Acanthia Cressa* from a mineral solution by a current of voltaic electricity, continued for eleven months. Mr. Gray, F.R.S., of the British Museum, who was present, doubted the possibility of their production, as similar experiments had been made by Mr. Children of the same establishment, but without any success.

FOUNTAIN OF FIRE.—I have just visited a lot of natural gas, which rises through a small river about 1½ mile from Pont-y-priid (Newbridge), Glamorganshire; the brook has its source in the valley between the Llantrisant and Dinas Mountains, flows into the Rhonda, and joins the Taff at Newbridge. In the centre of this river, or brook, is a continued rising, or strong bubble, of about eight or ten inches above the stream, as though the water was blown up by a current of air, or natural gas (and correspondingly cold to the hand)—this, when ignited, produces a powerful flame, from four to five feet in height, with intense heat; and the result is the same as burning sulphur. There are two jets in the field adjoining, and three on the other side the brook; but they are not so strong as the one passing through the water. The effect by night is grand and beautiful; having the appearance of one mass of large brilliant blue flames. This singular phenomenon was accidentally discovered by one of the men at the adjoining mill, about two months since. Not having yet seen any account of it in print, I have sent you some particulars collected on the spot.—Correspondent of the *Albion*.

SUPPRESSION OF THE RIVER NUNN.—At the usual monthly meeting of the Commissioners of the Birmingham Street Act, on Monday last, on the minute in reference to the subject of an inquiry as to the best means of effecting an abatement of the nuisance occasioned, having been read, Mr. Turner said that the committee were not prepared to make any report, but he was happy to inform the commissioners that the nuisance on long complained of in Birmingham, arising from the smoke of steam ferries, was in a fair way of being done away with. The patent of Mr. Williams (of which Mr. Dricks was the agent) had been tried at Mr. Clifford's mill, in Foxley-street, with the most complete success; and he believed that if the principle were generally adopted, the complaints in reference to this subject would not only be put to an end, but that a considerable saving would be effected by mill owners and manufacturers in the reduced consumption of fuel; he thought it was the duty of those commissioners who had forborne to give the plan a fair trial, and thus set an example to others in establishing a nuisance in Birmingham which had become almost insupportable.

LEAPFROG, AND HANCOCK'S WATER-WHEEL.—We understand that a company have just started a powerful engine at their new works, Lutterworth, Leicestershire, made by Mr. John Wigg, Newmarket, Fife. The engine, which is able to raise 50,000 gallons of water per day, is a fine specimen of workmanship, carrying out all the advantages of the expanding principle to the utmost extent, with a variety of improvements for economy of fuel; the "water" took place in the production of several of the leading engineers of the time, and was completely successful, and fully maintained the high reputation of the establishment from which it originated.

MEETINGS OF SCIENTIFIC BODIES.
IN THE MINING WEEK.

SOCIETY.	PLACE OF MEETING.	DAY.	HOUR.
Royal Botanical	Regent's-park	Saturday	4 P.M.
Royal Geographical	8, Waterloo-place	Monday	8 P.M.
British Architects	14, Grosvenor-street	Monday	8 P.M.
Medical	Bolt-court, Fleet-street	Monday	8 P.M.
Statistical	4, St. Martin's-place	Tuesday	8 P.M.
Linnean	St. James's-square	Tuesday	8 P.M.
Horological	21, Regent-street	Tuesday	8 P.M.
Civil Engineers	25, Great George-street	Tuesday	8 P.M.
London Electrical	Adelaide-street	Tuesday	8 P.M.
Chemical	47, Leicester-square	Tuesday	8 P.M.
Society of Arts	Adelphi	Wednesday	7 1/2 P.M.
London Institution	Finchley-circus	Wednesday	8 P.M.
Microscopical	21, Regent-street	Wednesday	8 P.M.
Royal	Brompton House	Thursday	8 P.M.
Antiquaries	Brompton House	Thursday	8 P.M.
Royal Institution	Albemarle-street	Friday	8 P.M.
Botanical	39, Bedford-street, Cav. St.	Friday	8 P.M.
Royal Asiatic	14, Grafton-street	Saturday	2 P.M.
Westminster Medical	Kent Hall	Saturday	8 P.M.
Mathematical	Cripley-street, Spitalfields	Saturday	8 P.M.

PUBLIC COMPANIES.

COMPANY.	MEETING.
London Bank	Goldsmith Coffee-house, March 14
West Durham Railway	George and Vulture Tavern, 16
Bleas Bridge Mining Association	George and Vulture Tavern, 17
Manchester and Leeds Railway	Office, Manchester, 17
London and Brighton Railway	London Tavern, 17
Aylesbury Railway Company	White Hart Inn, Aylesbury, 17
British Gas-Light Company	11, George-yard, Lombard-street, 23
St. Wh. Charlotte Mining Assoc.	George and Vulture Tavern, 29
Canada Company	Canada House, 29
West London Railway	11, Abchurch-lane, April 7
East Tretton Mining Company	34, March 14, 5, St. Mildred's-court.
British and Gloucester Railway	34, per sh. 31, Barnet, Home, and Co.
Northern Coal Mining Company	31, Barnet, Home, and Co.
Northern and Eastern Railway	31, Barnet, Home, and Co.
Hongkong & Shanghai S.S. Co.	31, Barnet, Home, and Co.
Miner's Association	31, Barnet, Home, and Co.
Miner's Company	31, Barnet, Home, and Co.
Irish Waste Land In. Society	31, Barnet, Home, and Co.
Canadian Iron & Steel Co.	31, Barnet, Home, and Co.

NOTICES TO CORRESPONDENTS.

PATENT DISC AND REGENERATING ENGINE.—We have received the report of Josiah Farley, Esq., C.E., which will receive early attention.

COMPARISON OF COALS.—We have to acknowledge receipt of the letter of our Manchester correspondent, which we must decline inserting, as containing no new facts, or, indeed, any feature of novelty, if we may except the personalities to which the writer indulges, amounting to nothing which has come under our notice. We regret personalities at all times, for they detract from the value of a communication, while abuse or invective can never fill up the vacuum created by the absence of argument or facts. We beg to assure our correspondent that the paragraph to which he refers, and which he supposes to have emanated from one of the parties interested in making known the success attributed to the invention, was framed from the report to which it alludes, and which, on a second perusal, appears to us to require no alteration. Another paragraph of a like nature will be found in our column of to-day. We are not in the habit of inserting paragraphs transmitted without giving the authority of a correspondent, and then only on the conviction that they are not manufactured to advance any sinister object.

FROM WHAT WE HAVE HEARD OF THE Llanidloes Patent Engine Company, we are desirous of becoming further acquainted with the patentee and the apparently economical method of applying water power which he has been the means of introducing. Will our correspondent favour us with some particulars to lay before our readers?

WEST COAST COMPANY.—Some remarks on the extraordinary decision of the House of Lords in this case are deferred until our next.

WE cannot insert the communication respecting the Westminster Loan and Investment Joint Stock Company—we know nothing whatever of the parties referred to, and have strong doubts as to the correctness of several of the assertions of our correspondent.

"N. V." (Hollywell).—Whether the desired information be received or not, we hope often to hear from our correspondent, to whose communications we shall at all times, we need hardly add, feel pleasure in paying the utmost attention.

"F. W. R." (York).—Probably in our next.

"H. M." (Redruth).—The papers referred to have never reached us—should our correspondent possess, or be enabled to obtain, duplicates, we should esteem it a great favour were he to forward them, as we are very desirous of having such valuable statistical information as must be contained in the documents mentioned.

LORD AUDLEY'S JOSEPH FINE, JOSEPH FINE & LORD AUDLEY.—We are obliged to our correspondent for the information conveyed as to the proceedings in the English and Irish Courts of Chancery between the noble lord and the ignominious. We understand Mr. Henry Staines is now employed by Lord Audley, having cast off his late client Joseph. This must be glorious things for the West Coast Mining Company and the lawyers. More litigation, more delay. We hope, in our case, for the sake of our national epiglottis, that Government will not be subjected to the cost of Joseph's travels in search of another Ballydoon on the other side the Atlantic—in other respects it might be profitable to the community.

Mr. Parkin's paper on Railway Engineering. Mr. John Phillips's on an Improvement in the Axle and Deceleration of Deep Wells. Mr. Donnan's Descriptive Observations on Various Subjects, and several miscellaneous articles, are unavoidably postponed.

YALE.—The letter of "B." came to hand, but too late for insertion in our present Number.

In consequence of the numerous applications made to the Editor on subject of Advertisements which have appeared in the columns of the MINING JOURNAL, with reference to articles or materials used in the working of mines and the construction of railways, arrangements have been partially effected, whereby all information necessary can be obtained on application of the office of the Journal, as also reference made to the various works, plans, drawings, and specifications, and where specimens may be seen, if being intended to derive a result in the shape of a paper or illustration. It is further arranged that the office of the Journal, in the case of being asked for the office of the MINING JOURNAL, the medium of conveying information on all matters connected with mineral property, where plans and particulars of mines and mining materials for disposal may be forwarded and obtained. Repertories given to the several mining districts will undertake drawings and furnish plans, sections, and reports, on mineral property and mining undertakings.

THE MINING JOURNAL,
Railway and Commercial Gazette.

LONDON, MARCH 12, 1843.

The financial statement submitted by Sir R. PEARCE, in the House of Commons last night, in the enumeration of the proposed new taxes, includes an export duty of 4s. per ton on all coal shipped, whether in British or foreign vessels. Heretofore this duty has been confined to coal exported in foreign ships, with the view of affording encouragement to the British shipping interest, on which no duty was imposed. The operation of the reciprocity duties, however, has been to exempt foreign ships from the duty, which it was originally intended to levy on the export of coals, and which was exemplified by the honourable baronet in reviewing the exports during the last ten years, the quantity exported rapidly increasing, while the duty received was in a much greater ratio diminishing.

In 1831 the quantity was	315,000 tons	Duty £20,250
" 1832 "	445,000 "	" 28,710
" 1833 "	1,125,000 "	" 72,500
" 1834 "	1,307,000 "	" 84,450

Thus it will be seen, that, while in 1833, a duty of 4s. 710d. was received on the exports of that year, this amount had, in the year 1834, dwindled down to 8000d., or a reduction of nine-tenths, while the quantity had increased threefold—so that, instead of a revenue which, based on the returns of 1833, should have been 105,000d., only 8000d. was actually received. The proposition of Government is, that in future all coal exported, whether in British or foreign vessels, shall be subjected to the duty of 4s. per ton, thereby, as Sir R. PEARCE observed, securing a revenue, from this source, of about 200,000d. a year—at the same time, operating to the encouragement of native industry—coal being necessary to manufactures, and contributing, as it does, by its export, to increase competition

abroad with our manufactures at home, and thus rendering a most important material of our own industry a great assistance to that of rival nations.

The papers relating to the sulphur question, and the claims of British subjects on Sicily, moved for by Dr. BOWRING in the House of Commons, were laid on the table of the House on Thursday last, from which we learn the exact terms on which the negotiation has been concluded—a mixed commission having been appointed to liquidate the claims of British subjects arising out of the sulphur monopoly. The claims laid before the committee appear to have amounted to 65,610d. 5s. 5d., whereas the awards are about one-third of that claimed, being 21,307d. 14s., with interest at 6 per cent. per annum, from the date of the claim to the period at which the payment may be effected, and which, it is presumed, will take place on an early day, his VOLCANIC MAJESTY having (it is said) "caused arrangements to be made for paying off the sums awarded with as little delay as possible," although not bound to do so, by the agreement entered into, for twelve months from the date of the commission being closed. The following comprise the heads of agreement entered into between the two Governments, defining those who should be entitled to compensation:—

1. Those parties who, having become mine proprietors or lessees previously to the 9th of July, 1838, the date of the contract with TAYLOR & CO., may have experienced impediments in the extraction or exportation of their sulphur, and may, in consequence of those impediments, have suffered losses duly substantiated.

2. Those who, previously to the said period, having entered into contracts for the delivery of sulphur, may have found it impossible to fulfil their engagements, or may have been deprived of the profits stipulated for on their transactions.

3. Those who, having bought sulphur, the exportation of which had been either prohibited or limited, or subjected to more onerous conditions, may have, in consequence, sustained losses capable of positive proof.

The report of the commissioners proceeds to state the course they adopted for ascertaining the correctness of the claims made, and the grounds on which they made their award—expressing their conviction "that the claimants have been fully indemnified for such of their losses as can, strictly speaking, be attributed to the monopoly contract, which was abolished in July, 1840."

Such is the sum and substance of the papers submitted to the House, involving a question which rendered, at one time, questionable whether other proceedings than those to which a commission of inquiry is confined would not have taken place, while the important position in which the mine adventurers and mining interest of this country is placed, has been entirely overlooked or neglected. We have so oft invited attention to the subject, that any observations we may make can only be a repetition of the oft told tale; but we cannot allow the present opportunity to pass without again expressing the opinions we entertain, and, moreover, endeavouring to impress on those who are embarked in sulphur mines the duty they owe to themselves of bringing immediately under the consideration of Government the injustice done to the adventurous capitalist, by allowing his Majesty the KING of the Two SICILIES to get out of the mess in which he had involved himself by the mere payment of 21,000d.—and that not by way of reparation to the consumer or the shipping interest, who were the principal sufferers, but to the British claimants, who were proprietors of sulphur mines in Sicily. Why, the very circumstance of raising the question whether he should compensate those who introduced capital into his country, to employ and feed the population, is in itself sufficient evidence of the dishonesty of the man.

Those who have recovered from the Neapolitan Government, it will be seen, are parties whose claims were but of a secondary nature—they have been allowed contemplated profits with 6 per cent. interest—they have been satisfied not for any losses they might have sustained from their operations or the fluctuating state of the English or foreign markets, but have had an award made in their favour for imaginary profits which they might have realised—and yet the consumers of England, taking 35,000 tons per annum, which in two years alone would make a difference of 400,000d. in the cost of the article rendered in this country, to that which was its value before the monopoly of TAYLOR & CO.—are left unnoticed. If we look further into the question as to its bearings on the mining interest, we shall find that to the enterprise of the miner the consumer is more indebted for the reduction of the price in the article than to any act of the Government, who appear to have confined their interference to the protection of the Sicilian, and not the home miner. Upon the price of Sicilian sulphur being raised to 13d. or 14d. per ton, consequent on the monopoly complained of, attention was drawn to the vast resources of this country in the production of an article of commerce from the otherwise refuse area, or such as were found to be too strongly charged with sulphur, and containing too small a proportion of metallic products to pay the expense of working at the price obtained for sulphur antecedent to that period. The advances made in science also aided the employment of sulphur ores or pyrites in lieu of Sicilian sulphur, and hence the article became of commercial value, and operations on an extended scale, more especially in the county of Wicklow, carried on with a spirit of enterprise which has been ill repaid. Confining ourselves, for the moment, to this particular district, we have only again to observe that it can produce a quantity of ore equal to the imports of foreign sulphur, and which can be rendered to the consumer at a price far below the latter article, if that protection be only afforded by our Government at home. The consequence of the mines being opened to an extent so as to secure to the consumer his supply at a reasonable cost (say, 24s. to 25s. per ton, delivered at Wicklow), proved the value to be attached to our home mines compared with those of his Sicilian MAJESTY, who thereupon having sold his stock for ready money (at less than one-half the market value previously obtained for the article)—expressed his resolve to throw off the duty of 11.13s., or thereabouts, and which he announced should take place in January, 1842. The result of these dishonest moves on the part of his MAJESTY has been, that the price of sulphur ore which would pay at 20s. per ton at the mines, or 25s. put on board at Wicklow, and consequently standing the consumers at Runcorn, St. Helen's, and other immediate localities, at 31s. to 32s. per ton, were reduced, and now are at least 20 to 25 per cent. under the price previously obtained, to which the miner was justly entitled—as affording him remunerative returns for the capital employed—while he conferred a benefit on the consumer, and secured him from the effects of foreign monopoly.

We could well fill our columns with arguments and facts in support of our case, but if the mining interest itself will not take up the question, how can it be supposed, with influence such as they possess, that any act or advocacy of ours can effect any real good? Our object, therefore, is rather to direct their attention to their own interests, leaving them to pursue the course they may think most fitting, while our columns will ever be open to discussion on the subject, and our best efforts afforded to the enforcement of justice being done to the English capitalist who embarks his property in one home mine, or, at least, to his being placed on an equality with the English capitalist who places his money in those of foreign mines, more especially as in the present case, where the Government had not even the honesty to afford protection to those who had been the means of developing the resources of the country and employing the population.

The following is a list of claims, according to the Parliamentary returns, presented to the commissioners for losses sustained by British

subjects, from being impeded in the production or exportation of their sulphur by the monopoly, showing the quantities charged for and allowed, and the several sums awarded by the commission—the numbers 1, 2, and 3, in the first column, apply to the several grounds on which the award was made, as noticed in the preceding remarks:—

Class.	Claimants.	Charged.*	Allowed.*	Total claims. English (ex. 87).	Total award. Engl. (ex. 87).
1.	G. Wood and Co.	122,478	105,382	117,380 0 10	45,482 2 1
1.	G. Wood and Co.	105,847	84,678	11,849 19 9	3804 14 9
1.	G. Wood and Co.	9,636	7,716	1,849 13 6	648 15 0
1.	G. Wood and Co.	7,243	7,243	8,482 19 0	1127 17 11
2.	G. Wood and Co.	2,842	2,842	1,277 15 0	421 1 1
1.	Morrison and Co.	28,391	12,915	1,274 1 0	308 13 9
1.	Morrison and Co.	4,508	4,508	1,915 11 11	741 0 1
1.	Morrison and Co.	3,208	3,208	706 18 0	284 11 3
1.	Morrison and Co.	3,810	3,810	260 9 0	231 19 7
2.	Morrison and Co.	3,810	3,810	789 13 6	702 2 1
2.	Morrison and Co.	3,096	—	807 16 8	—
3.	Morrison and Co.	2,716	2,716	670 9 2	319 4 10
3.	Price, Turner, and Co.	21,807	16,606	10,248 15 4	3508 25 6
3.	W. Leaf and Co.	26,136	26,136	17,731 0 4	1678 19 2
3.	Frank Hall	2,500	2,500	1,172 18 6	283 9 7
3.	F. and R. Sanderson	2,500	2,500	821 4 1	821 4 7
2.	Mathey, Oats, and Co.	700	700	211 3 0	32 12 0
3.	Thurbern Rose	671	671	158 12 0	41 0 0
2.	Samuel Lowell	6,700	—	1,180 0 0	—
Total claims		286,420	286,140	455,510 5 5	231,307 14 0

* For quantities of sulphur (in cantars), the production or export of which was impeded.

The letter of a correspondent, which appears in our columns of to-day, impugns the accuracy of our statements with reference to the cost of Sicilian sulphur—whose letter is also deserving of note, as holding out encouragement to the home miner. We are glad, at all times, to be set right, for it is hardly to be expected that we can be in possession of all detail connected with the varied subjects which form matter of Editorial observation and comment, whether as regards the mineral productions of "the bowels of the earth" or the legal productions of the "ermine on the Bench"—whether discussing railways or banks—in exposing fraud—or lending our humble aid in the development of our mineral resources—and the application of chemical and metallurgical science, not to omit the more important science of geology, or that of practical engineering. We do not profess to be possessed of more than ordinary acquaintance with those subjects on which we are led to express an opinion, but we invariably avoid committing ourselves, without possessing something like accurate data, or such, at least, as may be expected to pass current.

Such was the course observed as regards the comparison between foreign and home sulphur, and we are disposed to adhere to our own figures until our correspondent is prepared to give us facts as well as figures. Mr. LEIGHTON says, he considers the price quoted by us, put on board, as "very much too high," and accordingly, without giving any grounds on which he would enable us to form an opinion, he fixes the price in Sicily at 2d. 5s., instead of 4d.—the latter being our quotation. We presume that our correspondent is aware of the distance of the mines from the shipping port—the cost of extraction, and shipping and commission charges, which, if he would add together, he will find far exceed 2d. 5s. per ton. Again, as to the duty, he is there in error, as he will discover if he refers to calculations made in former Numbers of the Journal, taken from official returns and documents; and then, again, on subject of freight, we differ a trifle. Now, without entering into a discussion whether we or our correspondent are right or wrong—for, on that subject we shall be glad to compare notes, through the medium of the post—we will at once proceed to notice his letter on the more important point, that of the sulphur mines of this country being able to compete with those of Sicily, which he assumes, for argument sake, can be imported at 4d. 10s. per ton, including the import duty, but allowing the export duty to be withdrawn.

We will not enter into the question how far the British sulphur could compete with this reduction in price, if effected, for our correspondent does not furnish us with any data on which to arrive at an opinion; yet he asserts, and we are willing to take his assertions as facts, which are to be determined by the result of scientific inquiries and practical tests, that "sulphur, in a state of the greatest purity, can be obtained from sulphur ore, mundie, or pyrites, by a cheap and easy process, which being carried on at the mines, all extra charges would be saved." We do not propose following our correspondent in his theory, or the practice adopted in smelting copper ores, while we may draw attention to his statement, that "at present a quantity of sulphur is required in smelting copper," which process, he continues in his opinion, may "be gradually improved, and, indeed, ultimately so modified as to be accomplished without the use of sulphur," and further adds, that if works were established for commencing the manufacture of sulphur, a quantity of highly oxidised metals would be obtained.

There can be no doubt but that the theory advanced by Mr. LEIGHTON is well worthy the consideration of the sulphur miner, or the copper miner, where his ores are of low produce, and strongly impregnated with sulphur, and, under any circumstances, whether "a flux quite as effective, and not so expensive or injurious in its use as a compound of sulphur and iron" in the smelting of copper ores, can be obtained, as asserted by our correspondent, or not, is a matter of little moment to the sulphur miner, as he would have the means of making more of his produce by an extended market for the article. We wish some of our able correspondents, interested in the subject, would take it up, and avail themselves of our columns in drawing attention to its importance.

IMPROVED STEAM-ENGINE.—(From a Correspondent.)—Messrs. Remond have just completed the erection of a steam-engine on the Hornblower and Woolf principle—two cylinders—one working with high pressure steam as it issues from the boiler, and the other a low pressure, using the steam as it issues from the cylinder, which it then expands five times, consuming only between 2 lbs. and 3 lbs. of fuel per horse-power, or about one-third of the usual quantity. The qualities of this engine have been tested in the presence of several scientific men, when it was found that on a trial of eleven working hours it consumed at the rate of 132 lbs. of coal per hour, the engine then exerting an effective working power of sixty horse; this was at the rate of only 2 1/2 lbs. per horse-power per hour. The water required for condensation was only about one-third of the usual quantity. The duty, taken in Cornish measures of 94 lbs. of coal per bushel, was equal to 85,000,000 lbs. raised one foot, or upwards of 101,356,758 lbs. raised with one cwt. of coal; in the performance of this duty the engine averaged sixteen strokes per minute. It is a reciprocating rotary engine, and owes its great superiority to judicious proportions and superior workmanship.

INCURSION OF A MINING CAPTAIN.—At a meeting of magistrates, held in Truro, on Wednesday last, Captain William Thomas was charged with committing to assault on Benjamin Chivall. It appeared from the evidence that the defendant had been formerly engaged as captain at the North Wheal Towan Mine, but having been discharged by Mr. Backhouse, the owner, the assault originated in a disputed right on the part of Captain Thomas to prevent the removal of his share, he holding a slight interest or share in the mine. It also appeared that he had endeavored to assume the control of the mine subsequent to his discharge. The magistrates expressed their opinion that it would be most dangerous to the mining interests of Cornwall, to permit an agent, however he might alter be a shareholder, to set the manager and owner at defiance. The assault having been proved, Captain Thomas was convicted in the penalty of 10s. and costs, or, in default of payment, two months imprisonment. The fine and costs were paid.

DATA FOR THE USE OF BLAST-FURNACE MANAGERS.

BY SAMUEL BALDWIN ROBERTS, Esq.

(Mineral and Metallurgical Chemist, Nantyglo, Monmouthshire.)

5.—PREPARATIONS FOR SMELTING.

In my previous letters is stated an average of the materials required to add a ton of pig-iron, in this district of country, and with a cold-blast, and which, to save the trouble of repeatedly referring to such letters, I will here recapitulate:—1st. 4480 lbs. of coke, presumed to contain 8332 lbs. of pure carbon, and 448 lbs. of earths, or oxides. 2d. 3040 lbs. of calcined lime, containing 2240 lbs. of iron, 960 lbs. oxygen, and 1840 lbs. of foreign earths and oxides. 3d. 2240 lbs. of limestone, estimated to contain 1179 lbs. of caustic lime, 926.6 carbonic acid, 89.6 of silica, and 44.8 of water. 4th. 360,000 cubical feet, or 27,000 lbs. of atmospheric air, of which 5649 lbs. of oxygen, 19,771 nitrogen, 190 lbs. of water, in vapour, and 40 of carbonic acid, are supposed to be blown into the furnace, and 1350 lbs. are allowed for leakage of apparatus and loss of air & tyes, &c.

The atmospheric air, and all the oxygen, hydrogen, nitrogen, and carbon having previously been disposed of, we will now proceed to the separation of the lime, silica, and other earths and oxides, from the iron in the furnace; for, it should be particularly observed, that iron is an eluct, and not a product, of the operation in question, as some modern philosophers would tempt us to believe, except so far as it may be combined with carbon, or other alloying bodies; and good and bad are qualities governed entirely by the more or less admixture of foreign matters with pure iron, and not from any inherent properties it may possess—otherwise we should, on the detection of such properties, have proof of the existence of more than one element termed iron, and chemistry would, consequently, be obliged to remodel all its metallurgical philosophy. Previous, however, to entering upon the blast-furnace process, I will say a few words as to an improved shape and dimensions of such furnaces, and also give the form of analysis of several kinds of iron ore, limestone, &c., in order to show the best mode of proceeding for furnace managers to pursue with their respective materials; the actual analysis must, of course, be locally determined by the operatives themselves, who should also acquire an accurate knowledge of the action of the earths, and reaction of the compounds to be operated upon at known temperatures; for, under any other circumstances, the blast-furnace process will be subject to great and continual irregularities, and the results no better than, if not entirely, the effect of chance.

The furnaces here (i. e., in Monmouthshire and South Wales) are about forty-five feet in height, with single boshes, tapering from fourteen or sixteen feet diameter at their tops to five or six feet at their junction with the crucible; and the tunnel, or body, of the furnace is a truncated cone, of the same diameter as the top of the boshes, tapering in a height of twenty-eight or thirty feet, to five or six feet across at the charging plate. Instead of this construction, I would recommend a furnace of the following shape and dimensions, viz.:

1. The hearth to be five feet across, from side to side—the length and depth according to the fancy of the operator.

2. The crucible to be five feet square at the tyes, and six feet square at the top, in a height of six feet.

3. The lower bosh to spread from six feet square at bottom, to twelve feet diameter, in a height of four feet, producing an angle of about 54 degrees.

4. The lower body to be cylindrical for four feet in height and twelve feet diameter.

5. The upper bosh twelve feet diameter at bottom, and eighteen feet at top, in a height of four feet.

6. The upper body to be eighteen feet diameter, and cylindrical, to the height of six feet.

7. The tunnel, a truncated cone, eighteen feet diameter at bottom, and nine feet at top, and a height of twenty feet.

8. The charging-plate, chimney, or tunnel-head, at the pleasure of the operator.

With a furnace of this description, it is confidently imagined the materials would be less compressed about the tyes than they now are, and, consequently less resistance would be presented to the blast, which would then not regurgitate at the tyes to the extent it now does, and a better effect may certainly be expected. In furnaces of the present construction the first cause of their falling off in make and quality is, impediments collecting about the tyes, arising from the imperfect fusion of the materials under operation, and the pressure of an incumbent column of mine, coke, and limestone, to the amount, probably, of forty or fifty tons—the result often being a rocky mass of cooled cinder (in which are imbedded shots of iron, pieces of coke, and a mixture of raw and partially-fused mine and limestone), occupying nearly the whole area of the crucible, and to the depth of sometimes more than two feet from and above an horizontal section of the furnace at the tyes; in which case there is often one-third of the blast prevented from entering the furnace, and the temperature, consequently, becomes reduced to a degree inadequate to the production of any desirable result. The only advisable means of removing the impediments alluded to are, either to cut them out of the hearth and crucible by force, or to throw the furnace, by the aid of a very active flux, into a fit of scowering, until she has sufficiently cleared herself. One of the advantages of a hot-blast is, the ready and rapid fusion of the materials in the crucible of the furnace, thereby preventing the accumulation of such impediments as are above alluded to. Now, with regard to the analysis of materials for the use of the ironmaster, the contents of the mines, limestones, and coke ashes, should be tabulated somewhat thus—

TABLE OF CALCINED MINES.

100 parts, by weight, contain of	Iron.	Oxygen.	Silica.	Lime.	Alum.	Magn.	Total.
Red vein mine	42	18	12	1	1	—	100 parts
Black vein mine	48	17.5	12.5	1	1	—	100
Black shale	48	17.5	12.5	1	1	—	100
Black shale	48	17.5	12.5	1	1	—	100
Grey vein	48	17.5	12.5	1	1	—	100
Grey vein	48	17.5	12.5	1	1	—	100
Brown vein	48	17.5	12.5	1	1	—	100

[These names and quantities are necessarily assumed, and for two reasons—first, it would be impossible to give a table of the elements of mines, &c., that would suit for any two iron works; and, second, few ironmasters would be pleased with a public exposure of the value of their respective materials. These assumed tables will, however, answer every purpose as data for the smelter.]

TABLE OF LIMESTONES.

100 parts contain of	Lime.	Carb. acid.	Silica.	Alum.	Magn.	Ox. Iron.	Total.
White limestone	84.86	15.14	1	1	—	—	100 parts
Black limestone	84.86	15.14	1	1	—	—	100
Grey limestone	84.86	15.14	1	1	—	—	100
Black limestone	84.86	15.14	1	1	—	—	100

TABLE OF COKE ASHES.

100 parts contain of	Silica.	Alum.	Lime.	Carb. acid.	Magn.	Ox. Iron.	Total.
Coke ash from red vein mine	44	14	2.5	1.5	—	—	100 parts
Coke ash from black vein mine	44	14	2.5	1.5	—	—	100
Coke ash from grey vein mine	44	14	2.5	1.5	—	—	100
Coke ash from brown vein mine	44	14	2.5	1.5	—	—	100

As it may sometimes be necessary to use marble, shales, fire-clays, jacks (a hard kind of mine), and other materials, as adjusting fluxes to the blast-furnace, a table of their constituent elements, when prepared for use, will likewise be desirable for the guidance of the smelter; not, be it particularly observed, that a mere knowledge of their elements will be all that a good manager should rest satisfied with; for, unless he also makes himself fully acquainted with the reciprocal action of mineral bodies, but especially of silica, lime, alumina, and magnesia, at the usual temperature of blast-furnaces (say, a temperature generated by the consumption of nine tons of coke in twelve hours, or three burrows per hour), and obtains correct information as to the best proportions of the several elements for forming a clear and colorless glass, without the aid of oxide of iron, it

will be in vain for him to expect either good or uniform results, otherwise than as the effects of chance. This is a point which has scarcely ever been attended to in iron smelting, although, upon a due knowledge of which, the whole art of furnace-managing entirely depends; its solution is this—there is a very strong quiescent affinity existing between protoxide of iron and silica, which affinity can only effectually be overcome by bringing the siliceous oxide into actual contact (i. e., by fusion) with substances that have a stronger affinity for both the oxide and the silica than that which at the time may hold them in union. In the case in question the lime and alumina on the one hand, and the carbon of the fuel on the other, will, at a due temperature, readily effect the separation by a compound, or double decomposition, as it were; the lime and alumina, by attracting out and uniting with the silica, will instantly become a perfect and fluid glass, and the carbon of the fuel will, from the affinity existing between it and iron, at a high temperature, as instantly decompose the oxide (now left at liberty for its oxygen to enter into new combinations, in consequence of the silica being separated from it) and unite with the metallic iron, so as to produce a carburet, which, from being very fusible, and of great specific gravity, will fall through the glassy cinder into the bottom of the furnace, and be then safely protected from the injurious effects of the blast; under these circumstances, a furnace may be said to work well, and a good and uniform result may safely be depended upon.

There is also another subject to which an ironmaster will have to direct his particular attention, which is, the cinder produced from the mineral residue of his materials, for this is the index of all the chemical changes which take place in a smelting-furnace, and which changes often occur daily, and often from hour to hour. Now, furnace cinders are of various qualities, from a clear colorless fluid, and perfect glass, indicating a complete separation of the iron from the other materials, to a dense black scowering cinder, or scoria, containing a large portion of oxide of iron, and indicating a considerable waste of material, and, if long continued, bringing down sure destruction upon the furnace.

From the result of numerous analysis of furnace cinder, of all qualities and colours, from the very best kind to the very worst, it appears that, when the proportions of silica, lime, and alumina are in each other as 3, 2, and 1, the cinder will be clear and colorless, and the iron result soft and good, no impediment previously existing in the furnace; but should the earthy residuums consist of four parts silica, two lime, and one alumina, then the cinder will appear of a reddish, or foxy, colour, when drawn out thin, as it comes from the fall of the furnace, and looking through it; such a cinder will indicate a hard iron result, and the furnace (if so previously worked well) will have a tendency to fall still farther off as to quality; and should there be a larger portion of silica in the residuums, it would very soon put the furnace completely out of order, and would finally "gob her up," as it is technically termed, were not a reaction to take place, by the revived iron entangled in the cinder becoming again oxidized by the blast, so as readily to form a reunion with the excess of silica, which compound, acting as a powerful flux upon the imperfectly-fused materials in the crucible, will soon remove the accumulations from the furnace, in the shape of a black scowering cinder, as above-described; the cause of which accumulations being the infusibility of the residual earths, principally the silica in this district, at the existing temperature of the furnace, after the iron has been revived, and consequently separated therefrom. The remedy in this case will be an addition of such materials, or fluxes, as may increase the fusibility of the residuums, or else by augmenting the quantities of lime and alumina, or reducing the silica, until the proportions are—one alumina, two lime, and three silica, or as nearly so as possible—for, in these proportions, those elements readily fuse into a complete and perfect glass, without the aid of protoxide of iron; consequently, in forming his furnace charges, the smelter should refer to his analytical tables, in order to ascertain the quantities of silica and alumina (which are the principal residuums of the mines and fuels of this district of country) in the materials required to yield a ton of pig-iron; and should the alumina amount to one-third the weight of the silica, he will then only have to employ as much limestone as will yield, in caustic lime, twice the amount of alumina; the resulting cinder will then be a colorless compound, readily fusible, at the usual temperature of a blast-furnace, without the aid of protoxide of iron, and the process will be perfect and unchangeable, until one or other of the earths in question be reduced, or augmented, in quantity, so as materially to fall away from the relative proportions above enumerated. Should the alumina in the residuums not amount to one-third the weight of the silica, an adjusting flux, or mixture, should be employed to supply the deficiency—namely, a mine containing an excess of alumina, or else a due proportion of marl, fire-clay, shale, or jack; and if the silica should be deficient in quantity (in which case the cinder will be thick andropy, although, perhaps, of a good colour, and not flow readily from the furnace), mines, or other materials, yielding an excess of silica, should be employed, until its proportions are to the other residuums as above-stated. The quantity of lime to be employed in the process under consideration will necessarily depend upon the amount of silica and alumina in the mines, fuels, and adjusting fluxes to be operated upon in the furnace.

The composition of furnace mixtures will form the subject of my next communication.

ORIGINAL CORRESPONDENCE.

PRESSURE ENGINES & WATER-WHEELS.

TO THE EDITOR OF THE MINING JOURNAL.

SIR,—I beg to say that I have had the pleasure of personally knowing Mr. John Budge, of Callington, for many years past, and have had in my possession his valuable tables published in 1825, from that period, called the *Practical Miner's Guide*. These tables have been of great use to me, and I never discovered any inaccuracies in the calculations. On pages 86, 87, and 88, is a rule for discovering the power of a water-wheel, with the following example:—

Required the power of a water-wheel, the diameter being 46 feet, the buckets thirty inches long, 12 inches deep, and 6 inches wide, with 14 inch between each bucket, and the crank 3 feet long?
 240 buckets, each containing 78.13 lbs. water = 18,750 + 3 = 6250
 46 + 2 = 23 = 20 + 3 = 6.66 power, of the lever
 6250 × 6.66 = 41,625 lbs.
 Deduct for friction one-fifth..... 8,325 ..
 33,300 lbs.

will draw a 12-inch lift of pumps 113 fathoms nearly, 44-horse power.

By same rule—What duty will 1000 feet of water per minute perform over a water-wheel 50 feet high, crank 5 feet, to work 10-foot strokes in shaft eight strokes per minute, or 125 feet water for each revolution?

125 + 2 = 6.25 = 12.5 = 50
 Deduct one-third..... 16.66—33.33

Deduct friction one-fifth..... 6.66—26.66, which comes to the following result, viz.—125 feet over a 50-foot wheel will raise 26.66 feet same height 50 feet. One 100-foot, or two 50-foot wheels, will raise 53.33 feet 50 feet high.

A 50-foot pressure—125 feet per stroke
 Deduct 30.5 feet .. 75—50

Deduct one-fifth..... 10—40 feet water fifty feet high.

A 100-foot pressure—125 feet per stroke
 Deduct 30.5 feet..... 37.5—87.5

Deduct one-fifth..... 17.5—70 feet 100 feet high, or 140 feet fifty feet high.—The following examples prove:—

1st.—As 26.66; 100 :: 40; Answer 150, or 50 per cent. superior.

2d.—As 53.33; 100 :: 140 :: 262, or 140 ..

Allowing Mr. Budge to be right in his wheel calculations, and that my pressure calculations are not overrated, then we will find that a 50-foot pressure is 50 per cent. better than a 50-foot wheel, and a 100-foot pressure is 162 per cent. better than a 100-foot wheel.

Cardigan, March 5. F. V. W.

THE CONSTRUCTION AND APPLICATION OF WATER-WHEELS.

TO THE EDITOR OF THE MINING JOURNAL.

SIR,—Your correspondent, "N. V.," Holywell, seems to have overlooked Messrs. Whitwell and Sturges's patent hydraulic machines, of which much has been said and written of late. "N. V." should state the use intended to be made of his 50-foot fall, and also the situation, on which much depends in erecting water-wheels and pressure-engines; if for pumping, and the situation be favourable for a pressure-engine, this may be attended with the least friction and expense; if a rotary-engine be required, and the situation be favourable for a wheel, then the wheel is best; if a 50-foot wheel be applied to pump, there are many advantages in using tooth-gear.

Bangor, March 5.

THE CONSTRUCTION AND APPLICATION OF WATER-WHEELS.

TO THE EDITOR OF THE MINING JOURNAL.

SIR,—Your paper of the 26th of February I have not seen till to-day. I am sorry my writing should have been so unintelligible as to have led to any error. I believe, however, it is so far intelligible to practical men, that they will have no difficulty in understanding it, or answering the question I have proposed. The wheel I mentioned in this county making thirty revolutions per minute is twelve feet diameter; it lifts a hammer with a cam ring under the point. In comparing the power of a water-wheel with a pressure-engine, I have said the crank travels three times through the length of stroke, and is effective twice: I should have said with one crank, for with two cranks, as proposed by your Bickhigh correspondent, the wheel is effective through the whole revolution. The one I mentioned in this county working with teeth and pinion has the cranks at right angles.

Holywell, Flintshire, March 2. —

WATER-WHEELS.

TO THE EDITOR OF THE MINING JOURNAL.

SIR,—Among the numerous readers of your valuable paper, perhaps some one of them will devote five minutes of their time to assist me in the following difficulty:—I am entrusted with the management of a lead mine, which has not been worked for twenty years, and, being limited to expense, have to contend with the following difficulties in the best manner I can. The old work is eighteen fathoms deep, and it is now proposed to sink down the shaft ten fathoms deeper; the feed of water is great, and I have only the water-wheel, belonging to the old work, to clear the water; this wheel is cast-iron, 25 ft. diameter and 6 ft. at the breast, and is well supplied with water; the feet is brought about six feet over the highest part of the wheel. My fear is that this power is scarcely sufficient to clear the water, and I am desirous to find out the best sized cylinder for the pumps to do this work best. I propose dividing her work into two lifts—the upper lift to be from the bottom of the old work—viz., thirty-six yards, and another lift to sink the work twenty yards. Some of the old workmen say that the working-piece of the pump before used was eighteen inches diameter. I shall feel much obliged to any practical gentlemen informing me what diameter he would recommend me to use in the working-piece to each lift with this power, and any other hint he may be able to suggest, by a reply in your valuable paper.

Cardigan, March 9.

PRESENT STATE OF THE IRON TRADE.

TO THE EDITOR OF THE MINING JOURNAL.

SIR,—Having seen a statement in your paper of the 5th inst. of the different blast-furnaces in England and Scotland, and at same time requesting information of any errors that may have been stated, I, therefore, beg to correct it, as to my works; you have it thus—

Proprietors.	Name of Works.	No. of fur.	For, in blast
W. YoungOmoa.....	1

and it should be as under, viz. —

Robert StewartOmoa.....	3
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Of the works having been purchased by me upwards of two years ago. Of the two furnaces out of blast, one is quite ready, and would have been blowing, had it not been for the arrangement as to reduction of make, and the other is a good furnace, although out of repair.

Glasgow Iron Works, Glasgow, March 9. R. STEWART.

ON THE NATURE AND QUALITIES OF SULPHUR.

TO THE EDITOR OF THE MINING JOURNAL.

SIR,—I am happy in observing that you still devote attention to the interests of the proprietors of mines yielding sulphur ore. I look upon the subject as one of great national importance, and offering a fine field for speculation. I fear I may be considered presumptuous in calling in question the accuracy of any statement which may appear as a leading article in your deservedly-esteemed Journal, but as I flatter myself I am well acquainted with sulphur, both as an article of commerce and as a chemical agent, I take the liberty of stating that you have estimated the cost of Sicilian sulphur imported into this country very much too high in your leading article of the 19th ult.—you there state it at 86. 8s. per ton. I consider the cost in Sicily much too high at 45. per ton; and I think the export duty is only 17. 10s., instead of 17. 15s. With the present superabundance of shipping, I should say, too, that 17. 15s. was above an average freight—say 17. 10s. In making any estimate of the cost of sulphur, we must not rely upon the permanency of any impost which His Majesty the King of Sicily may be disposed to levy; it is better, therefore, to calculate upon a duty free export from Sicily. About fifteen years since I bought a large parcel of Sicilian sulphur, duty paid, in England at 47. 10s. per ton. I believe it has not been so low since, and it might have been then sold at a loss. I have since paid as high as 165. per ton.

I am satisfied that the United Kingdom may be made independent of foreign countries for its supply of sulphur, its minerals containing a superabundance—in fact, more sulphur is wasted annually than would be sufficient to supply all the wants of our manufacturers. The manufacture of sulphur would be an excellent speculation, even taking the price at a minimum—say 47. 10s.—apportioned thus—Price in Sicily, 27. 8s.; freight, 17. 10s.; duty in England, 10s.; charges, 5s. The greatest consumption of sulphur is for the manufacture of sulphuric acid, for bleaching compositions, and the decomposition of salt for soda and alkali. Sulphuric acid may certainly be prepared direct from the ores in their natural state, but it is attended with much additional expense, from the extra bulk of materials to be moved about, and the impossibility of obtaining the whole, nor anything like, of the real per centage of sulphur which the ores contain. Sulphur in a state of the greatest purity can be obtained from sulphur ore, mundie, or pyrites, by a cheap and easy process, which being carried out at the mines all extra charges would be saved. Were not the copper smelters such out-and-out Conservatives, this process might be conducted with great advantage in connection with copper works, when an adequate supply of sulphur could be procured from that quarter. The waste of sulphur from copper works occasions an immensity of damage, as any one can bear witness to who has travelled from Neath to Swansea. It certainly used to grieve me much at the time this country was on the verge of a war with the kingdom of the Two Sicilies, to reflect that such an amount of damage should be caused by the waste and destruction of so large a quantity of the very article (sulphur) as would supply all the wants of our manufacturers, about which the Governments of the two countries were embroiled in such an angry dispute. At present a quantity of sulphur is required in smelting copper. The present process is termed by scientific men "a barbarous process," but it really is not when theoretically examined into. The different ores from which copper is smelted are very numerous, and the combinations of the copper, and the substances with which it is associated, are as various. There is no one flux that has as yet been found to act on generally, or act on universally, in decomposing these various combinations, and amalgamating such a diversity of substances into one common mass, copper slag, as proto-sulphuret of iron. A large quantity of this is used as the flux, and is prepared by calcining a large proportion of pure sulphury copper ore, by which a portion of sulphur is dissipated, and the iron (which constitutes a considerable item in such ore) is partially oxidized. When a general mixture of the ore is melted with this calcined ore, the proto-sulphuret of iron fuses all the various substances, and dissolves the copper which settles to the bottom.

The impurities, or slag, being then skimmed off, the copper is left in an impure state, combined with a portion of sulphur, which it is difficult to get rid of, but is effected by an alternating series of calcinations and fusions, requiring about six distinct operations. In the calcining process the surfaces of the pieces of metal become oxidized; in the fusion the oxygen of the oxidized surface combines with the sulphur of the interior, forming sulphurous acid, which is expelled by the high temperature, and thus by sufficient repetitions of the processes the copper becomes pure enough for refining, or tapping, as the workmen term it. Such appears to me to be the theory of the present process of smelting copper, which may be gradually improved, and, indeed, ultimately so modified, as to be accomplished without the use of sulphur. If a company could be formed to commence the manufacture of sulphur, a quantity of highly oxidized metals would be prepared. I have long entertained the idea, that if a portion of such oxidized ore were to be mixed with fresh sulphury ore, and a general mixture of the ore to be smelted, there being all fused together, the same results would be obtained as if the whole of the sulphury ore had been calcined—thus offering to the manufacturer of sulphur a market for his refuse, and saving the copper smelter the entire process of calcination. This is the first improvement in copper smelting which I should attempt; at the same time, I confidently state my con-

Admission that an alternating process of oxidation and decomposition may be applied to the melted copper in the furnace, that it may be drawn off from the furnace sufficiently pure by a single fusion to be fit for the refinery. While upon the subject I will state further, that I could devise a flux quite as effective, and not so expensive, or injurious in its use, as a compound of sulphur and iron.

I remain, Sir, your's, &c.,
T. H. LEITCHMAN.
(These remarks on this communication will be found in another column.)

ON THE COMBUSTION AND CHEMICAL PROPERTIES OF COAL.

TO THE EDITOR OF THE MINING JOURNAL.

Sir,—*Fuel is a very good maxim* when not carried to an extreme. I fear, however, that the progress of the discussion on the properties of coal must appear rather too slow to most persons, who, like myself, might have expected it would have been brought to a close before this time. Mr. Williams, in his last letter, acknowledges himself to be in error, and I will, therefore, just remind him how the subject at present stands. My letter of the 20th January, "On the Combustion of Smoke," and of 9th February, "On the Use of Hot Air in Furnaces," are yet unanswered, and we have had no more corrections of my paper, "On the Chemical Properties of Coal," since the 25th December last. I should be sorry, however, to hurry Mr. Williams, if he finds any difficulty in bringing the artillery of his arguments to bear upon these subjects, and I will, therefore, in the meantime, make a few observations on his letter in your Journal of the 5th instant.

Regarding the disputed passage in Mr. Williams's letter of 1st January, as to the same quantity of air being required at all periods of the combustion of the coal, we agree, as to the expression used, and only differ in the meaning which they are supposed to convey. Now, although Mr. Williams must, of course, know what he really intended to express—and I am bound to take his present reading of the passage to be what he originally intended—still I maintain that the passage did not express the meaning he attributes to it. The statement, that "there is no such time, and there are no such circumstances," when "a regulating air admission valve would be necessary," was evidence, I conceived, that Mr. Williams considered there ought to be an equal volume of air admitted at all stages of the combustion; for, if this were not the case, a regulating air admission valve would certainly be required. But Mr. Williams denies there is any such time, or any such circumstances, as would require a diminished influx of air; and, therefore, what I stated must be the necessary consequence—viz., that he considers "the same quantity of air is required to consume the gaseous products of coal at all periods of combustion." The new reading of this passage, that there is no time when there is no demand for a supply of air to the burning fuel, is unquestionably correct as to the fact. I am much obliged to Mr. Williams for his invitation to inspect his experiments at Liverpool, but it is unnecessary for me to do so, because I am perfectly acquainted with the fact which he states. Indeed, I should have conceived no person at all acquainted with the theory or the practice of combustion could be ignorant of the fact, that a considerable supply of air is at all times required for effecting perfect combustion of every description of fuel. Mr. Williams states this is a common oversight, that it appears almost natural; but I can only say, I never met with any instance where such an oversight existed, and which would only proceed from the most perfect ignorance of the subject. But, what I asserted (and what, also, Mr. Williams appears now to assert) was, that less air was required for the combustion of the coal after the hydro-carburets were evolved than was required for the combustion of these hydro-carburets; and, if there be no regulating valve to the furnace, I do not see how this condition can be fulfilled, although I quite agree that there are practical difficulties in the application of such valves, which are not easy to overcome.

As regards Mr. Williams's remarks, that I have wholly omitted any notice of the distinction which he draws between the use of hot air to coke and anthracite, and its application to combustible gases, I have to observe, I answer to his inquiry, that the omission was intentional, and not accidental. My views of the advantages of hot air to the combustion of anthracite have been already fully stated in my paper "On the Constitution of Coal," and, as Mr. Williams's views coincided with mine, I conceived there was nothing required to be said on this subject. But I beg to set him right on one subject upon which he has fallen into an error. He considers I have, in all my arguments, "confounded two processes, an essentially different as the combustion of solid and gaseous bodies." Now, I beg to assure him, I have not done so. He will not find any one single instance where I have confounded these processes, nor will he find, as he asserts, that Sir H. Davy's experiments will bear out his arguments, and I approve mine, respecting the use of hot air. Such statements, without even an attempt at proving them, will not, I think, be considered as a very satisfactory way of conducting a discussion of this sort. This much, however, I am ready to grant him—that, whether the air or the gas be heated, is a matter of but little consequence, so long as the mean temperature of the air and the gas, when mixed, shall be sufficiently high for the mixture to enter into combustion. But, as the air constitutes the larger volume, it is certain that the gas must be greatly lowered in temperature by the admission of so large a body of cold air as is required for the combustion of the gas; and, frequently, this is the case to a very injurious extent.

Yours, &c.,
Charles HUGHES.

"DOINGS" IN THE NORTH.

TO THE EDITOR OF THE MINING JOURNAL.

Sir,—You have taken an interest in the Durham County Coal Company, and we are much indebted to you for your independent and valuable assistance. It is to be hoped that we have now not only "scotched the snake," but "killed it," and that what you may hereafter have to chronicle, respecting this company, will be of dividends, premiums, and bonuses. There is no matter, however, to which I wish to draw your attention, as a preliminary to these happy days, and that is, the suit which the company are engaged in with the Stockton and Darlington Railway Company. Be it first understood that the railway company is mainly under the direction of Quakers; it is not needful to enter into the merits of the case between the two companies, which are well understood—enough it to say, that the railway company has been beaten in every court, and that the unanimous decision of the Judges is against them; but, in spite of this, holding, as the fruit of their opposition, about 10,000 tons of the funds of the Durham Coal Company, they seem determined to carry the case to the House of Lords, the shares of which are being furnished by the coal company's shareholders, however, the executive of the railway company follow such conduct, there are not wanting propellers—upright men—who shrink from their proceedings, and I could name one venerable man, who, upon the temporary advice which the law's delay affords, has entered his solemn protest against the attempt longer to withhold from our company that which the law has awarded to it. It is said that corporate bodies have corporate consciences, and, whilst I freely allow that the directors of this railway are individually honorable men, I must say that their conduct is deserving of the severest reprobation. Let them beware, and let the public indignation which has overtaken the originators of the company do not fall upon those who, by every obstruction which the law affords, are endeavoring to delay, and defeat, the claims of justice. The Quakers who have protested against this act of the railway company is himself a Quaker, and was once nearly the largest proprietor in the colliery, and still holds several shares; his judgment is, therefore, a disinterested one, and he calls upon the railway company to disgorge their ill-gotten gains, as it is plainly between them and one day's rest, and their further opposition can only be imputed to hatred and envy of the coal company, and to get off the evil day to an distant as possible. There are, no doubt, quibbles and subterfuges by which these gentlemen may postpone their consciences, but that which they would be ashamed to do individually, it is too base to do collectively; behind the shield of a public company, they may hope to escape a personal challenge, but the time may come, when it may be smelted to by upon each of these persons the full weight of his own responsibility in this matter.

Yours, &c.,
Charles HUGHES.

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DESTRUCTION OF BUCKET LEATHERS IN PUMPING ENGINES.

TO THE EDITOR OF THE MINING JOURNAL.

Sir,—As some of your able correspondents have suggested a remedy to "A Subscriber" (in Journal No. 346), for the prevention of the rapid destruction of bucket leathers in pumping-engines, where the water, as in some coal mines, causes the leather to decay in a week or ten days, I think the most charitable construction we can put on the silence observed on this subject, is, that no remedy has been discovered or become known. I would, therefore, humbly beg to suggest to "A Subscriber," to convert his drawing pumps into forcing pumps, and so bucket leather will be required. If not trespassing too much on your valuable columns, I may as well state my own experience in similar circumstances as that alluded to above. In a coal mine, where I happened to be employed a few years ago, at a new winning, the engine-shaft was only about thirty fathoms deep, but the quantity of water was excessive; by hard work, and close attention, the shaft was sunk with one set of pump trees, 10-inch working barrel, but as soon as the coal was struck, a second set had to be put down, and before much ground was opened, a third set, all one size, 11-inch pipes and 10-inch barrel, all working side by side, pumping from bottom to surface. Now, if any two out of the three sets could have been kept in perfect order, and in constant work, they would have kept the pit dry, but from the rapid destruction of bucket leathers, this was found to be impossible; two men had always to be in readiness to go down to change buckets, as soon as any of them should show defect, which, in general, was two or three times a week; and, while change was making in one set, all three should be idle, and water rising in the shaft, so that the delays, men's hire, and cost of leather, became intolerable. As a remedy, a "plunger" (a forcing pump) was put down, and, from henceforth, all went on smoothly. Clinks, mounted with double leathers, "haik-mouth" fashion (which is far preferable to "butterfly" mounting) did not require to be changed for months. In consequence, the working of the colliery soon became profitable, which it never would by any other mode of pumping, owing to delays and extra costs. I have sometimes thought that the metallic nature of the water far exceed the colliers of the north, in the better arrangement of their pumping-engines and pump work, as the latter do the former in the superior ventilation of their mines. It might be well if they would be advised to take a lesson from each other, but this suggestion, I fear, neither will take in very good part; for, in whatever locality one goes into, each and all think (and will tell you) themselves the pinks of perfection in everything—so that, as was lately, and truly said, by an intelligent writer, in your Journal—"Fraydon tends to render the manager in one district unwilling to receive instruction from one of a different practice;" and, whether you go north or go west, we have all got the same song—"there is no place like home," "there is no place like home." A WORKMAN.

March 8.

[We enclose in the wish of our correspondent, that parties would adopt the principle of benefiting themselves by the experience of their neighbors, and, by the same rule, allow their neighbors the advantage of their own practical observation. "Give and take" is a motto which might be adopted, as a principle of action, by many others than "A Workman." We see no reason why any jealousy or rival feeling should exist on the part of practical, or any other body of men; the only rivalry should be—that of proving who most can benefit their fellow-men, by the promulgation of knowledge.]

PUBLIC UNDERTAKINGS AND PUBLIC CHARACTERS.

TO THE EDITOR OF THE MINING JOURNAL.

Sir,—May we not join in the cry of the dissentients from the Government, and exclaim "What will become of the country?" for surely it is a question which may be justly asked, when scarcely a week passes but we find some one of the early patrons of philanthropic objects, or grand undertakings, projected in this famous city, become, as it were, "marked men," or declared unable or unworthy to proceed in the good cause of benevolence or of nationality. Mr. W. Milnes Thomas, and his various suggestions for benefitting our mining districts, and, in so doing (though some evil-minded persons doubt their possibility), making the fortunes of his associates, are now almost forgotten—and so, indeed, were those of another enlightened gentleman, Mr. Joseph Pike, and perhaps might have remained so, but for the disinterestedness of a set of disinterested, calling themselves the West Coast Mining Company—who, I am informed, presume to question the legality, or propriety, of the decision of the first common and equity law lords in this or any other country, because, fourthly, it proclaimed his conduct as fair and honest—but let that pass. Recently has that worthy man, Alderman Thomas Wood, had time to recover his equanimity from the violent attack which he has sustained for an endeavour to benefit the poor of Dublin under a law which, by enabling them to procure cheap coal, &c., (and what, Sir, may I ask, is there in modern history to equal his forbearance, while supposed by his slanderous persecutors to be swarting from the provoking manner in which his truly charitable intentions had been misconstrued?)—when we find his friend, Mr. Thomas Lamm Murray, subject to similar indignities, and ejected from the direction of the Irish Waste Land Improvement Company, of which he was one of the earliest patrons (and to whom, indeed, we are, perhaps, more indebted than to Mr. Alderman Thomas Wood, for, as I am informed, he was the projector of several of the most useful, if not the most prosperous, undertakings of the day—among which may be named the National Loan Fund Assurance Company, the Languish Iron and Coal Company, the Metropolitan Patent Wood Paving Company, &c.). Again, we have Mr. Daniel O'Connell, too, held up to the world as a defender in that same undertaking, which, it is well known he has laboured so hard and so well to serve, because, in so doing he was furthering the dearest object of his heart, that of benefitting his beloved country and her distressed population; why, the blood of every Irishman must rise, and his indignation be exalted, at the illiberal reflections cast upon that patriotic man's conduct—but to whom, as of course he is well able, I leave the justification of his character, and the rehabilitation of his name. And last, though not least, it grieves me further to notice, that at the meeting of the Thames Tunnel Company, on Tuesday last, the directors announced that that spirited gentleman, Mr. Frederick Boucher—even he, Sir, who has done so much for our colonies, in the (as was thought) successful establishment of that, among other, great schemes, the British and Australasian Bank—was disqualified for the appointment of auditor. Oh! Mr. Editor, when we find the best interests of so many good and righteous men sacrificed to the petty jealousies and revengeful malice of a parcel of idle newspapers, for the sake, as I believe, of a little temporary popularity, is it not time to ask—"What will become of the country?" A FRIEND TO OUR BROTHERHOOD.

Thursford, 2d, March 2.

[Our correspondent surely has a disposition "to poke his fun" at our editor and ourselves, in signing himself "A Friend to our Brotherhood," if that be comprehended in the list the virtuous William Milnes Thomas, the late Joseph Pike, the immortal Alderman Thomas Wood, the respectable Mr. Thomas Lamm Murray, the defender Daniel O'Connell, Esq., Lord Mayor of Dublin, or that "spirited gentleman," celebrated Minister Frederick Boucher. All these names have in their time graced the public, and are each, in their to-day, obtaining the rewards they so justly merit. Each, and all of them, are cherished and respected from all honest and decent society;—as we may except Mr. Alderman Thomas Wood—who, by virtue of his acknowledged, is qualified to take his seat on the same bench with Alderman Pike, Thompson, Lamm, Campbell, and other members of the Court, and who, with all patience Michaelmas day, when the value of the Livestock of London will be pronounced whether in their opinion the worthy legal gentleman possesses more moral principle of honest virtue than the worthy or legal Alderman Milnes. Well may our independent ask, with "so many good and righteous men," "What will become of the country?" We reply in an appeal to the country's laws, and the expression of honest feeling on the part of the community at large, whenever the opportunity presents itself to record their opinions.]

CARBOUR.—A patent has recently been taken out for a new description of fuel under the above title, which is thus produced—A quantity of coal, charcoal, coke, brown coal, or peat, being reduced to a fine powder, in part into wooden chips and mixed with oil; two and a half parts of water, or one of clay or lime, are then added, and the mass worked into cakes by hand or machinery; these cakes being afterwards gradually dried, by the application of fire heat, become nearly as hard as stone, and in burning are said to give out more heat than any other known combustible. Should a fuel capable of producing a still more intense heat be required, oil and shell are combined with powdered coke and coal; the proportions of the ingredients for these quantities of combustion are as follows:—I. Twenty-four parts coal, six or seven of clay-water, one of oil, and one of lime. II. Twenty-four parts coal, six or seven of clay-water, and two of oil. III. Twenty-four coal, thirty-one of clay-water, and

GEOLOGICAL SOCIETY OF LONDON.

At the meeting of this society, on the 19th January (Mr. Marchant, President, in the chair), a paper "On the Recession of the Falls of Niagara," by Mr. Lyell, was read.

Professor Eaton published, in 1824, a correct section of the rocks between Lewistown and the Falls of Niagara; and in 1835, 1831, and 1833, Mr. Bakewell, Mr. De la Beche, and Mr. D. Rogers, laid before the public accounts of the phenomena of the Falls, and the physical structure and geology of the district. In the Report of the Geology of Western New York, for Mr. Mr. Conrad first declared his opinion, that all the formations of that country belonged to the Silurian series; but Mr. Lyell says, that the true geological succession of the rocks between Lewistown and the Falls was never well understood, until Mr. James Hall, the States geologist, published his report in 1838. After these allusions to previous labours, Mr. Lyell proceeds to give a brief account of the strata composing the Niagara district, and chiefly either from the reports of Mr. Hall, or from information obtained from that gentleman, while travelling with him during the autumn of 1838. The strata between Lakes Erie and Ontario are considered to belong to the middle and lower portions of the English Silurian system, and are divided into five principal formations—viz., 1. the Helderberg limestone; 2. the Onondago salt group; 3. the Niagara group; 4. the Protean group; and 5. the Ontario group. The first, or newest, constituting the country adjacent to Lake Erie, and called the Helderberg limestone, is considered, on account of its organic contents, to represent the Wenlock rocks of Mr. Murchison's Silurian system; and the correctness of this conclusion Mr. Lyell has verified by a personal examination of the strata, from the coal-field of Pennsylvania to the group in question, the intermediate formations containing organic remains which agree with those found in the Devonian system and Llandovery of England. In this part of the state of New York, and still further west, in Upper Canada, the Helderberg series is only fifty feet thick, but at Schenectady, 360 miles to the eastward, it attains a thickness of 300 feet. The Onondago salt group differs essentially from any member of the English Silurian system, consisting, with the exception of a stratum of limestone, the top, of red and green marls, with beds of gypsum, which are undistinguishable from the new red marls of England. The deposit is also non-siliceous. No rock salt has yet been found in the group, but lime occurs in frequent occurrence. On the line of the Niagara, the strata have been denuded, and are much concealed by overlying drift, but the thickness is estimated at not less than 800 feet, and Mr. Hall conceives, that, in some parts of New York, it is full 1000 feet. 3. The Niagara group first appears on approaching the rapids above the great cataract. It consists in the upper part of the Niagara or Lockport limestone, 120 feet thick, and in the lower part the Niagara or Rochester shale, which is eighty feet thick, and both contain fossils identical with those of the Wenlock series of England, but are dated with others peculiar to North America. The limestone constitutes the whole of the platform from the rapids to the escarpment at Lewistown, where its thickness is only thirty feet, and it rests persistently on the Niagara Rochester shale, which maintains throughout the same vertical direction. 4. The Protean group, which crops out at the base of the Falls, consists of a series of twenty-five feet of hard limestone, resting upon about four feet of shale; but at Rochester, in the Genesee River, it is better developed, and includes a bed of dark shale, with graptolites, and another of limestone, full of *Pentamerus oblongus* and *P. laticus*, considered by Mr. Conrad to be one species. 5. About a mile below the Falls the Ontario group begins to rise beneath the Protean, and extends to the escarpment at Queenstown, Lewistown, where its thickness is 200 feet, but to this dimension must be added 150 feet of inferior beds, composing the district between the escarpment and Lake Ontario. The group consists, in descending order, of several feet of red marl, with beds of hard sandstone in the upper part, twenty feet of hard white quartzite sandstone, and 250 feet of red marl and limestone. Mr. Lyell is of opinion, from a comparison of English Coralline and *Lamella fenestra* with those found in the groups 4 and 5, that the Protean and Ontario series represent the lower Silurian rocks of Great Britain. The dip of the whole of these groups is at a small angle to the south, and as the strike has been traced 150 miles to the eastward, and for a greater distance to the westward, the sections along the banks of the Niagara afford a key to the structure of a large portion of the State of New York. Beneath the groups just described, the lowest of which extends to Lake Ontario, another called the Mohawk group, exposed on the Canada side of Lake Ontario, and Mr. Lyell is of opinion that it belongs to a series of beds older than the lower Silurian rocks of England. The author then enters upon some details respecting the geographical distribution of the formations, the physical features of the country, particularly those presented by the platforms, composed of the Helderberg and the Niagara limestones, and the escarpments at their terminations north of Buffalo, and at Queenstown, to distance from the point where the Niagara flows out of Lake Erie to the Falls is sixteen miles, from the Falls to the Queenstown escarpment is one mile, and thence to Lake Ontario is also about seven miles. For the fifteen and a half miles, or from Lake Erie to the commencement of the platform, the descent of the river does not exceed fifteen feet, but in the next five miles, or to the edge of the cataract, it is forty-five feet; the perpendicular height of the Falls is 164 feet, and from their base to Queenstown, the descent of the river is about 300 feet, but thence to Lake Ontario not more than four feet. These measurements, Mr. Lyell says, are of importance in speculating on the past or future recession of the Falls. If the cataract were ever at Queenstown, its height must have been then twice what it is at the present time, and the descent of the river would have been double the present descent. The change in the level of the river between the Falls and Queenstown, with respect to the origin of this escarpment, the author shows that it cannot be assigned to a fault, the strata composing it, and extending from its base, preserving the same relative position as at Lockport or Rochester, and is of opinion, that it is due entirely to denudation, at a period when the strata extended to its foot. He is also convinced, that the Helderberg escarpment was likewise formed by the action of the sea. Mr. Lyell then enters upon the great question, whether the ravine through which the Niagara flows has been cut by the river, or was excavated by the same agent which produced the escarpment. His own observations have induced him to conclude that the latter is the case, and that the drainage of Lake Erie was at one time affected by a body of water flowing along a shallow valley which occupied the present line of the ravine, and agreed in character with the valleys or depressions, through which the Niagara now runs between Lake Erie and the Falls. Mr. Lyell assigns the following reasons for his conclusion—1. The breadth of the ravine being at the top only from 400 to 600 yards, and at the bottom from 200 to 400, between Queenstown and the Whirlpool. 2. From the fact that the river flows everywhere cut down in the regular notch. 3. From the fact that the Falls are now slowly receding. 4. From the existence of the remains of a freshwater deposit on Goat Island, and its depression at the top of the cliffs, half a mile lower down on both sides of the river, the signs of which accumulation he assigns to the body of water which flowed along the shallow valley in question. The objection which has been advanced against the inference, that the river has cut back the ravine and founded upon an indentation in the cliff, called the "Devil's Hair," between the Whirlpool and Queenstown, Mr. Lyell is of opinion is not valid as he observes that the river which now flows down the notch, aided by atmospheric agency, would be able to form the hollow. He alludes likewise to another indentation, noticed by himself and Mr. Hall, on the Canada side of the river, and near the whirlpool, the characters of which had appeared escaped previous observers.

Mr. Lyell does not attach much importance to the precise numerical estimates respecting the recession of the Falls during the last half century, but he notices the great changes which took place in 1818 and 1820, and others which have occurred within the memory of persons residing in the district; he mentions likewise a work published by a French missionary, Father Hennepin, containing a view of the Falls in 1678, and which, in addition to the two existing cascades, represents a third on the Canada side, crossing the Horse-shoe cataract at right angles. This cascade is also alluded to by a Swedish botanist Kalm, who published an account of the Falls in 1735; at that time it did not exist. Mr. Lyell then details the characters of the freshwater deposits on Goat Island and at the top of the cliffs, proving that it had been known previous to Mr. Bakewell's account of the Falls, and Mr. Hall has described it in his report for 1838. The deposit consists of marl, gravel, and sand, containing some species of fossil shells, all which still inhabit the Niagara. On the north-west extremity of Goat Island, twenty-four feet thick. On the right bank of the Niagara, opposite to Lewistown, a terrace twelve feet in altitude has been excavated by this recession, which is there about twenty-four feet in depth; and in digging, millions of years since, a mass of the *Strophomena americana* was discovered with the same species of *Strophomena* shells; a similar terrace is distinctly seen on the Canada side of the river, and at about the same level, the characters have not been investigated. These deposits demonstrate, Mr. Lyell says, the former position of the river at a level corresponding to that of the present recession of the cataract, and one half a mile below that point the river has receded, and the present recession of the Falls is due to the fact that the river has produced that terraced condition of the water necessary for the falling water, and he is of opinion that it cannot be shown the position of the river has not been cut back in any three miles. If this be admitted, as Mr. Lyell is inclined to do, the recession of the Falls would be produced by the action of the river, and he adds, if the waters continue to cut their way back, the recession of the Falls will be partially lost day, and the recession will be continuing with equal features similar to those of the Great Sand cataract. Assuming that the Falls were once at Queenstown, Mr. Lyell details the differences which must have occurred in the rate of recession, in the

